

ANTIBACTERIAL EFFICIENCY OF MODAL FABRIC TREATED WITH OXIDES OF Ti/Si/Zn NANOCOMPOSITES

A. R. Arul¹, T. E. Manjulavalli^{2*}, R. Venckatesh³ and G. Rajkumar⁴

¹Department of Physics, Kumaraguru College of Technology, Coimbatore-641 049, India

²Department of Physics, NGM College, Pollachi- 642001, India

³Department of Chemistry, Government Arts College, Udumalpet-642126, India

⁴Department of Fashion Technology, Kumaraguru College of Technology,
Coimbatore-641 049, India

*E-mail : emanjhu@gmail.com

ABSTRACT

The antibacterial efficiency of nano TiO₂, TiO₂/SiO₂, TiO₂/ZnO and TiO₂/SiO₂/ZnO Nanocomposites coated on modal fabric using dip coating was evaluated using Fourier infrared spectroscopy, Transmission electron microscopy, scanning electron microscopy and zone of inhibition against *Bacillus subtilis*. The particle size varies between 17.9 nm to 46.4 nm. The fabric treated with TiO₂/SiO₂/ZnO exhibited a maximum zone of inhibition of 53 mm followed by 47 mm and 43 mm for TiO₂ and TiO₂/ZnO treated fabrics respectively. TiO₂/SiO₂ treated fabric has shown a minimal zone of inhibition of 36 mm.

Keywords: Antibacterial efficiency, Nanocomposite, SEM, Modal fabric, FTIR

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INTRODUCTION

Controlling for harmful effects created by microorganisms is an important task in the present scenario for the healthier living of human beings. Natural equilibrium with the human body and living environments has a broad range of microorganisms, but a rapid and uncontrolled fast thriving of microbes can lead to some serious problems.¹⁻³

Clothing is one of the basic needs of human beings and different kinds of textile fibers are used for human clothing where comfort is the prime factor. Textiles provide a suitable substrate to grow microorganisms, especially at appropriate humidity and temperature, in contact with human body. The microorganisms have made three undesirable effects on textiles. The first includes degradation phenomena like coloring, staining, and determination of fibers and the second, producing an unpleasant odor, and finally, the third effect is an increase of potential health risks.⁴⁻⁶ Hence, it becomes essential that the textile materials should have some form of antibacterial protection imbibed in them if they are designed to use for longer period of time. Application of inorganic nanoparticles and their nanocomposites on textile materials have given a wide opportunity for antimicrobial and multifunctional modification of textiles.⁷ In most of the research materials, TiO₂, SiO₂, and ZnO nanoparticles have been used to achieve antibacterial properties in textiles. An enhancement in the antibacterial property was found in the modified nano TiO₂ treated cotton fabrics.⁸ In the antimicrobial treatment of cotton fabrics by TiO₂ nanoparticles; smaller nanoparticles have shown better results.⁹ Nano SiO₂ treated textiles have provided better microbiological controls in hospital interiors.¹⁰ A significant result was observed in the zinc oxide nanoparticle treated cotton fabrics against *S.Aureus*.¹¹ In the ZnO/TiO₂ composite treated nylon/cotton blended fabric, the highest antibacterial property against *E.coli* and *S.Aureus* was obtained at a concentration of 60 ppm.¹²

Many researchers have worked on the antibacterial modification of textiles on natural fibers such as cotton and wool, cotton/polyester, nylon/cotton.¹³⁻¹⁷ No literature is found on the antibacterial finishing of modal fabric. Modal is a regenerated cellulosic fiber and second generation viscose fiber. It is made out of pure wooden chips from the beech tree. High wet strength and extra softness are the distinguishing properties of

modal. It has excellent drape, good hygroscopic properties, good moisture regain and air permeability. Modal has 50% more hygroscopic per unit volume than cotton.¹⁸ Therefore, it is considered best for exercising clothing and health suit. Hence, this research work aims at the application and characterization of nano TiO₂, TiO₂/SiO₂, TiO₂/ZnO and TiO₂/SiO₂/ZnO nanocomposites coated on modal fabric using dip coating technique for the antibacterial efficacy.

EXPERIMENTAL

Materials and reagents

The bleached single jersey modal fabric was fabricated and the geometrical parameters of the fabric are given in Table-1. The chemicals used for the synthesis of TiO₂, TiO₂/SiO₂, TiO₂/ZnO and TiO₂/SiO₂/ZnO are TTIP (Titanium tetra isopropoxide), hydrochloric acid, ethanol, silicic acid, THF (Tetrahydrofuran) and ZnCl₂ which were procured from S.d.Fine chemicals, India. Lissapaln and acrylic binder were sourced from M/s Jeeva Chemicals, Tirupur, India. *Bacillus subtilis* (Gram-positive) bacteria was used to ascertain the antibacterial activity of nanoparticles coated modal fabric which was obtained from IMTEK, Chandigarh, India.

Table-1: Geometrical parameters of modal fabric

Material	Geometrical Parameters							
	Yarn Count (Ne)	Course /inch	Wales /inch	Stitch Density	Areal Density (g/m ²)	Thickness (mm)	Loop length (cm)	Tightness factor
Modal Fabric	26	34	23	782	184	0.49	0.46	10.23

Synthesis of TiO₂ nanoparticles

Sol-gel (Wet chemical) technique was adopted for the synthesis of TiO₂ nano particles. TTIP (Titanium tetra isopropoxide), hydrochloric acid and ethanol were used as a precursor, peptizing agent, and solvent medium respectively. TTIP was added to the ratio of 1:4:2 to a mixture containing hydrochloric acid and ethanol. The mixture was stirred for 1 hour at room temperature. Then 50 ml of distilled water was added to it and the temperature was raised to 50°C and stirred for 3 hours until the solution changed into a colorless gel. The high viscous gel was dried at room temperature to obtain a fine powder. The resulting powder was heated at 110° C for 1 hour in a hot air oven. Finally, the colorless powder was calcined at 600° C.¹⁹

Synthesis of TiO₂/SiO₂ nanocomposite

SiO₂ solution was prepared by mixing silicic acid with THF (Tetrahydrofuran) in the ratio of 1:2. SiO₂ sol was added drop-by-drop to the TiO₂ sol, which resulted in a yellowish brown solution. The mixture was stirred for 3 hours at room temperature, then the temperature was raised to 80° C, and stirring was continued for an hour. Finally, yellowish brown color changed to yellow, and then the solution was dried at room temperature. The final product was heated in hot air oven for about 110° C for 1 hour. Finally, all the composites were calcined at 600° C.¹⁹

Synthesis of TiO₂/ZnO nanocomposite

ZnO sol was prepared by dissolving ZnCl₂ in double distilled water. The mixture was stirred for 1 hour at 100° C. ZnO was added drop-by-drop to the TiO₂ sol, then the temperature was raised to 100° C, and stirring was continued for an hour. Finally, a colorless gel changed to grey, and then the solution was dried at room temperature. The powder thus obtained was heated in a hot air oven at about 110°C for 1 hour. Finally, all the composites were calcined at 600° C.

Synthesis of TiO₂/SiO₂/ZnO nanocomposite

ZnO sol was prepared using ZnCl₂ which was dissolved in double distilled water. The mixture was stirred for 1 hour and maintained at 110° C. ZnO was added drop-by-drop to the TiO₂ - SiO₂ sol, then the temperature was raised to 120°C, and stirring was continued for an hour. Finally, the yellow color changed

to light grey, and then the solution was dried at room temperature. Grey powder was then heated in hot air oven for about 110°C for 1 hour. Finally, all the composites were calcined at 600°C.

Antibacterial finishing of fabrics

0.2 g of nanomaterial was taken in a beaker with 10 ml distilled water. 0.5 ml of acrylic binder and 1 ml of surfactant were added to the same beaker. This solution was prepared at 30°C. Then the solution was sonicated for 15 min at a frequency of 50 Hz. Sonicated solution was immediately taken for coating in antibacterial finishing bath. The sample was treated by dip coating method. The modal fabric of size 3X3 cm was dipped in the solution for 10 min and padded using automatic paddler. The fabric was dried at 110°C for 4 minutes and cured at 150°C for 3 minutes in an oven. The same procedure was followed for all the four samples.

Surface Morphology Studies

The surface morphology of fabric samples treated with nanoparticles was examined by Carl Zeiss (Sigma), UK scanning electron microscope (SEM) with an accelerating voltage of 10 kV at x 2500 magnification and transmission electron microscope (TEM) with energy dispersive X-ray spectrometer (EDS) (JEOL JEM 2100).

Fourier infrared spectroscopy (FTIR)

Fourier transform infrared spectroscopy (Bruker, USA) was used to characterize the structures of TiO₂, TiO₂/SiO₂, TiO₂/ZnO and TiO₂/SiO₂/ZnO coated modal fabrics with a resolution of 4 cm⁻¹ in the range of 4000-400 cm⁻¹.

Antibacterial testing

The American Association of Textile Chemists and Colorists (AATCC) Test Method 147 is a qualitative antimicrobial test used to detect bacteriostatic activity on textile materials. This antimicrobial fabric and textile testing method are useful for obtaining a rough estimate of activity by the size of the zone of inhibition and the narrowing of the streaks caused by the presence of the antimicrobial agent permitting an estimate of the residual antimicrobial activity.

Determination of Zone of Inhibition (ZOI)

The required quantity of microbiological growth medium (Mueller Hinton Agar) was prepared and autoclaved at 120°C for 20 minutes. Approximately 10 to 20 ml of the media was poured into sterile Petri plate. It was allowed for solidification. Midlog phase culture of *Bacillus subtilis* was spread on MHA plates. Fabric sample of size 2 cm² was placed on the culture of Petri plate containing a pathogen. This was incubated for 24 to 48 hours at 37°C. The sample was taken and zone of inhibition was measured on all four sides using a scale and the mean values were recorded and compared with the untreated control specimen.

RESULTS AND DISCUSSION

Transmission Electron Micrography(TEM)

The Transmission electron micrographic images of the TiO₂, TiO₂/SiO₂, TiO₂/ZnO and TiO₂/SiO₂/ZnO nanoparticles are shown in Fig.-1. It is observed that most of the particle appears to be the spherical shape and rod-like morphology having a size between 17.9 nm to 46.4 nm. From TEM it is confirmed that the prepared particles are in nano range.

FTIR Spectra

The FTIR spectra of untreated fabric, TiO₂ treated, TiO₂/SiO₂ treated, TiO₂/ZnO treated and TiO₂/SiO₂/ZnO treated fabrics are depicted in Fig.-2. The absorption band at 1425 cm⁻¹ in Fig.-2(a) for the untreated fabric indicates the presence of both cellulose II and the amorphous cellulose which is the characteristic peak of the modal fiber.²⁰ The band at 1336 cm⁻¹ and 1312 cm⁻¹ are assigned to both crystalline celluloses

(Cel I and Cel II). Furthermore, another important characteristic that peak value 893 cm^{-1} confirm the absence of cellulose I in the modal fiber.²¹

In Figure-2(b), a strong band in the range of 900 cm^{-1} and 500 cm^{-1} was observed which is associated with the characteristic vibrational modes of TiO_2 . This confirms the formation of the TiO_2 mode in the modal fabric.⁹ A broad absorption peak at around 3332 cm^{-1} and 1643 cm^{-1} of molecular water was observed in the TiO_2 treated modal fabric.⁸

The peaks around 1100 cm^{-1} and 800 cm^{-1} correspond to symmetric and asymmetric stretching vibrations of Si-O-Si bonds respectively in Fig.- 2(c). The peak at 450 cm^{-1} indicates the bending mode of Si-O-Si. A small peak at 1017 cm^{-1} confirms the bonding between Ti and Si.²² A broad peak at 3347 cm^{-1} indicates OH stretching due to the presence of hydroxyl groups in the fiber.

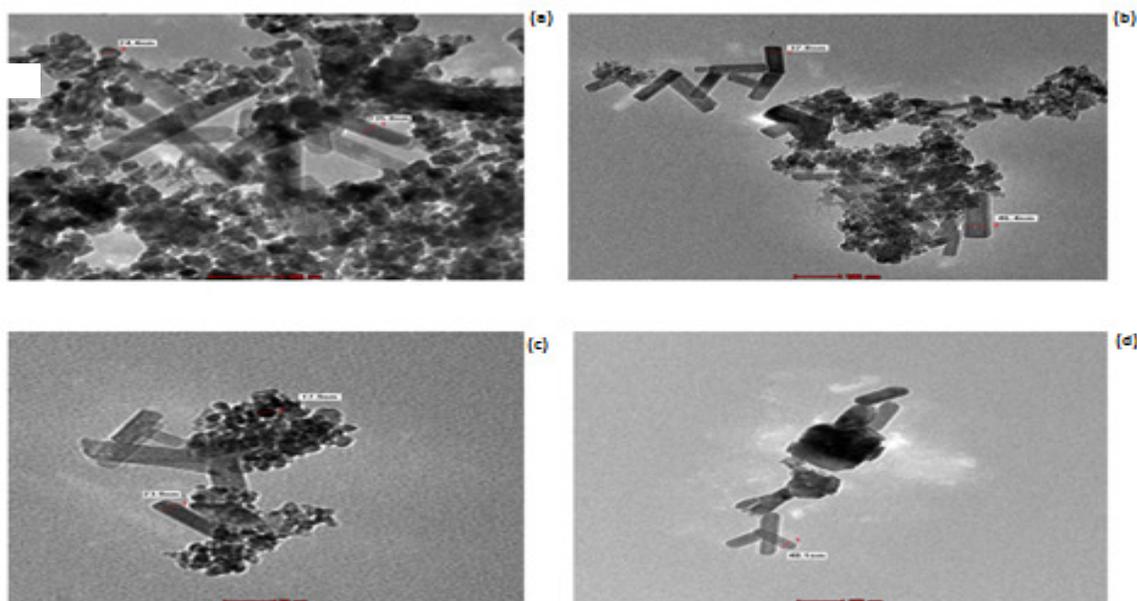


Fig.-1: Transmission electron microscopic images of the Nano composites (a) TiO_2 (b) $\text{TiO}_2/\text{SiO}_2$ (c) TiO_2/ZnO (d) $\text{TiO}_2/\text{SiO}_2/\text{ZnO}$

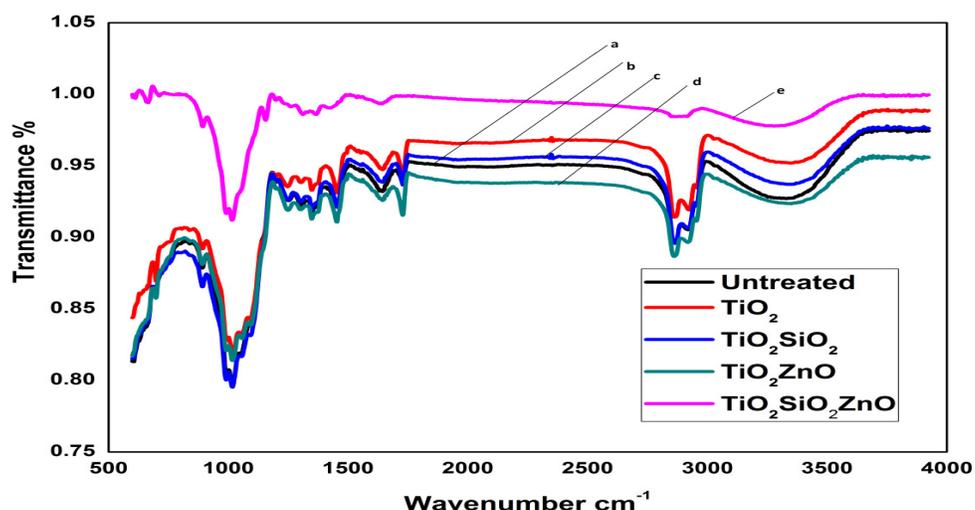


Fig.-2: Fourier transform infrared (FTIR) spectra of modal fabric(a) untreated fabric (b) Treated with TiO_2 (c) treated with $\text{TiO}_2/\text{SiO}_2$ (d) treated with TiO_2/ZnO (e) treated with $\text{TiO}_2/\text{SiO}_2/\text{ZnO}$ nanocomposites

In Figure-2(d), the peak in the range 891 cm^{-1} and 500 cm^{-1} is associated with the characteristic vibrational modes of TiO_2 . The peak at 450 cm^{-1} indicates the ZnO absorption band.²³ From this it is evident that both the TiO_2 and ZnO modes are present in the modal fabric.

In Figure 2(e), the characteristic peaks of all the three nanoparticles such as TiO_2 (891 cm^{-1} and 500 cm^{-1} , SiO_2 (982 cm^{-1}) and ZnO (432 cm^{-1}) are obtained in the spectrum.²⁴ This confirms the presence of all the three nanoparticles in the fabric.

Antibacterial efficacy

The results of disc diffusion tests in Fig.-3 shows the zone of inhibition of TiO_2 , $\text{TiO}_2/\text{SiO}_2$, TiO_2/ZnO and $\text{TiO}_2/\text{SiO}_2/\text{ZnO}$ treated modal fabrics for *Bacillus subtilis*. The growth of *Bacillus subtilis* around the untreated fabric is evident from the Fig.-3(a). The fabrics treated with nanoparticles have shown a reduction in the growth of the bacteria *Bacillus subtilis* around the fabric surface which is clearly observed in Fig.-3(b) to 3(e).

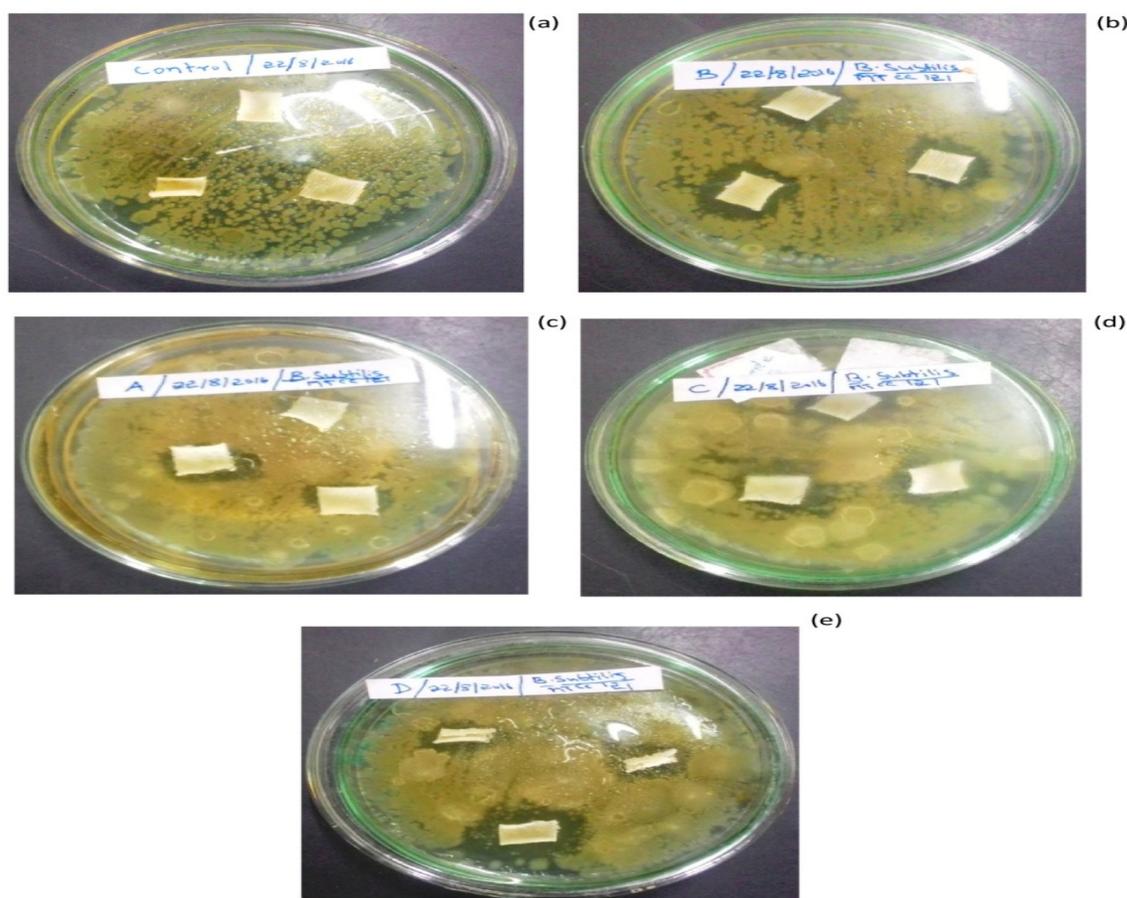


Fig.-3: Photographic images of Zone of inhibition of (a) untreated fabric (b) Treated with TiO_2 (c) treated with $\text{TiO}_2/\text{SiO}_2$ (d) treated With TiO_2/ZnO (e) treated with $\text{TiO}_2/\text{SiO}_2/\text{ZnO}$ nanocomposite

The fabric treated with $\text{TiO}_2/\text{SiO}_2/\text{ZnO}$ has provided a maximum zone of inhibition of 53 mm in Fig.-4 followed by TiO_2 treated sample (47 mm). This is due to the combined effect of $\text{TiO}_2/\text{SiO}_2/\text{ZnO}$ nanoparticles present in a single component. The sample treated with TiO_2/ZnO nanocomposite and $\text{TiO}_2/\text{SiO}_2$ nanocomposite have given a lower zone of inhibition of 43 mm and 36 mm respectively. The SEM micrographs of untreated, TiO_2 treated, $\text{TiO}_2/\text{SiO}_2$ treated, TiO_2/ZnO treated and $\text{TiO}_2/\text{SiO}_2/\text{ZnO}$ treated

- modal fabric are depicted in Fig.-5(a) to 5(e). The nano particles were not found in untreated modal fabric in Fig.-5(a).

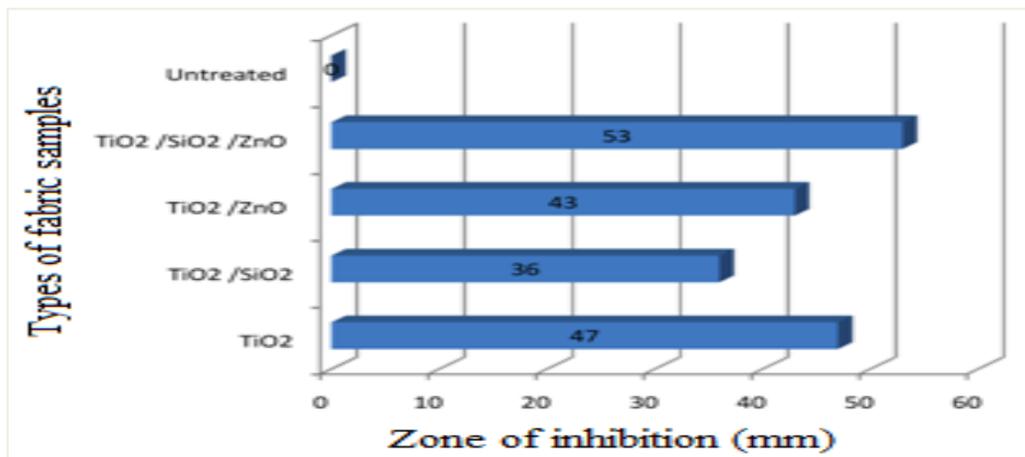


Fig.-4: Zone of inhibition

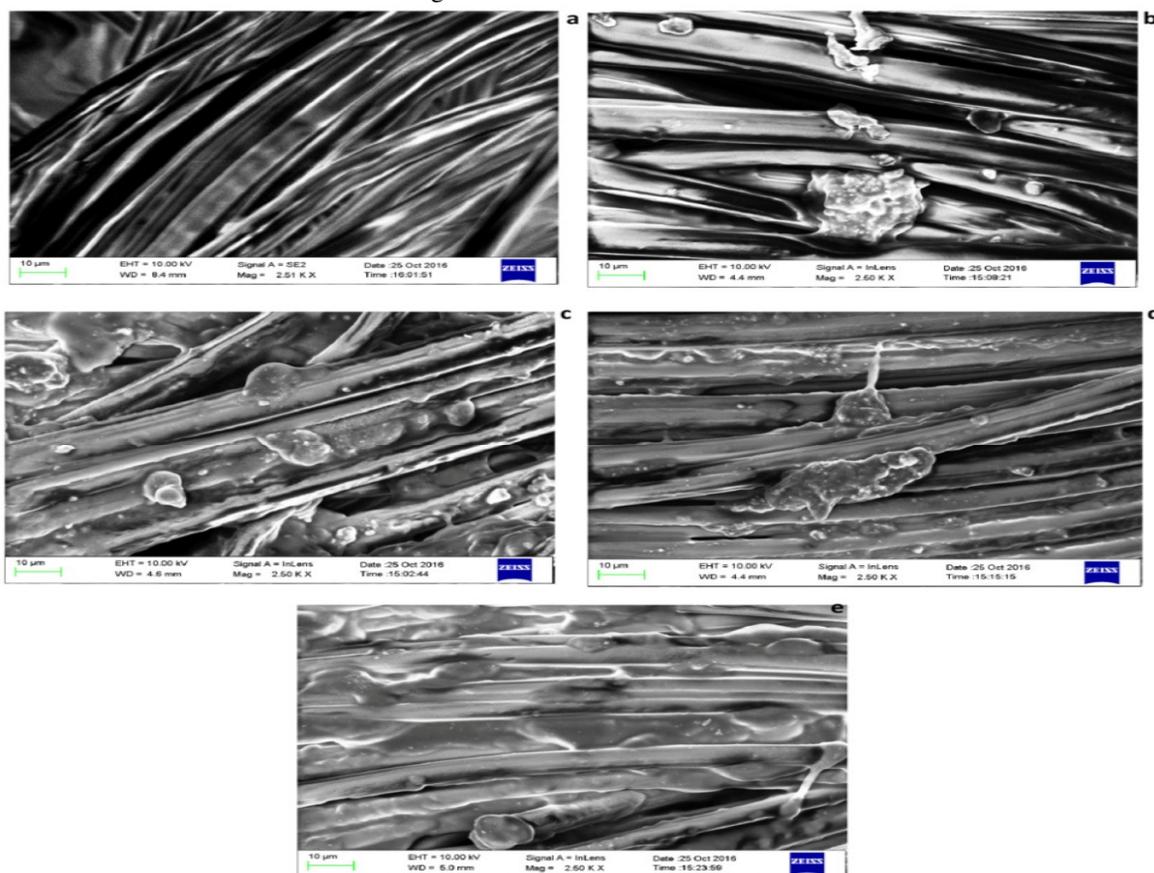


Fig.-5: Scanning electron microscope (SEM) images of (a) untreated fabric (b) Treated with TiO₂ (c) treated with TiO₂/SiO₂ (d) treated With TiO₂/ZnO (e) treated with TiO₂/SiO₂/ZnO nanocomposite

It is evident from the Fig.-5(b) to 5(e) that there are presences of nanoparticles on the modal fabric. The nanoparticles are well dispersed on the fiber surfaces and are finely embedded. An uneven and agglomerated patchy coating is found in the Fig.-5(b), 5(c) and 5(d). It is clearly observed in Fig.-5(e) that the TiO₂ /SiO₂ /ZnO nanocomposite particles were uniformly distributed over the fiber surface when

compared to Fig.-5(b), 5(c) and 5(d). This may be the reason for the enhanced antimicrobial efficacy when compared to other samples.

CONCLUSION

TiO₂ nanoparticles, TiO₂/SiO₂, TiO₂/ZnO and TiO₂/SiO₂/ZnO nanocomposites were prepared by the sol-gel method. These composites were applied to the modal fabric using dip coating method. The untreated and treated modal fabrics were characterized by SEM and FTIR. The antibacterial efficacy of treated fabrics was evaluated against *Bacillus subtilis* by measuring the zone of inhibition. It was observed from the study that the TiO₂/SiO₂/ZnO treated fabric has given a maximum zone of inhibition of 53 mm. The TiO₂ treated fabric has shown a zone of inhibition of 47 mm followed by 43 mm and 36 mm for TiO₂/ZnO and TiO₂/SiO₂ untreated and treated fabrics respectively.

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