

# A SYNTHESIS, CHARACTERIZATION AND ANTIBACTERIAL STUDIES OF COUMARINE DERIVATIVES SYNTHESIZED FROM 4-HYDROXY-CHROMEN-2-ONE AND COMPARISON WITH STANDARD DRUG

Islam Krasniqi<sup>1</sup> and Aziz Behrami<sup>2</sup>

<sup>1</sup>Faculty of Education, Public University of Prishtina, Prishtina-10030, Kosovo

<sup>2</sup>Public University of Mitrovica, PIM Trepca-40000, Mitrovica, Kosovo

\*E-mail: a.behrami\_chem@live.com

## ABSTRACT

In present paper, we report the organic syntheses of three compounds from 4-hydroxy-chromen-2-one and describe the results of antibacterial activity of purified compounds. 4-hydroxy-3-(3-Hydroxy-benzoyl)-chromen-2-one (**1a**), 3-(3-Hydroxy-benzoyl)-4-(2-hydroxy-phenylamino)-chromen-2-one (**2a**), N-[3-(3-Hydroxy-benzoyl)-2-oxo-7-thioureido-2H-chromen-4-yl]-N-(2-hydroxy-phenyl)-acetamide (**3a**), have been synthesized and characterized using melting points, IR spectra, <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectra. The antibacterial activity of synthesized compounds and streptomycin and cefalexine at concentrations of 2mg/ml, 3mg/ml and 5mg/ml, have been evaluated against three strains of bacterial culture; *Staphylococcus aureus*, *E.coli* and *Bacillus cereus*. The compounds show bacteriostatic and bactericidal activity.

**Keywords:** Coumarine derivatives, antibacterial activity, IR, <sup>1</sup>H-NM, <sup>13</sup>C-NMR, Streptomycin.

© RASAYAN. All rights reserved

## INTRODUCTION

Starting from 4-hydroxy-chromen-2-one (**a**); derivatives (**1a**, **2a**, **3a**) are synthesized. Coumarin derivatives are large group of heterocyclic with oxygen as heteroatom. Coumarin is a chemical compound (specifically, a benzo- $\alpha$ -pyrone) found in many plants notably in high concentration in the tonka bean (*Dipteryx odorata*), vanilla grass (*Anthoxanthum odoratum*), woodruff (*Galium odoratum*), mullein (*Verbascum* spp), and sweet grass (*Hierochloe odorata*). Coumarine and their derivatives have shown various biological<sup>1,2</sup> activities. Their fame has come mainly from their antithrombic<sup>3</sup>, antiinflammatory<sup>3</sup>, vasodilatory, and antiviral activities. With reflux and condensation we have synthesized some new coumarin derivatives and to investigate their antibacterial<sup>13</sup> activity against *Staphylococcus aureus*, *E.coli* and *Bacillus cereus*. The antibacterial activity of synthesized compounds is compared with antibacterial<sup>13</sup> activity of Cefalexine and Streptomycin.<sup>4,14</sup>

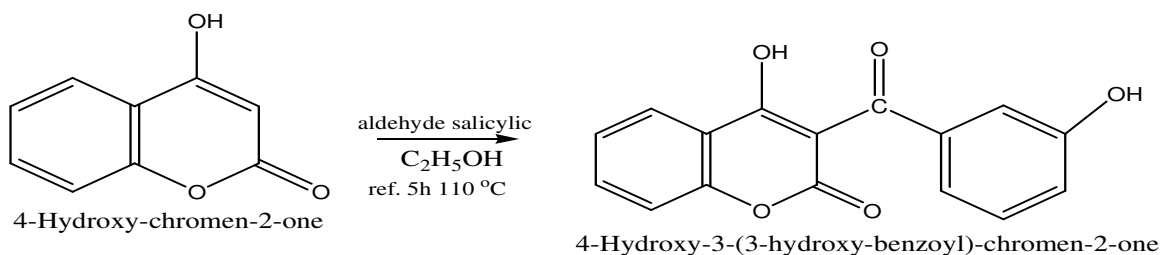
## EXPERIMENTAL

4-hydroxy-3-(3-Hydroxy-benzoyl)-chromen-2-one (**1a**), 3-(3-Hydroxy-benzoyl)-4-(2-hydroxy-phenylamino)-chromen-2-one (**2a**), N-[3-(3-Hydroxy-benzoyl)-2-oxo-7-thioureido-2H-chromen-4-yl]-N-(2-hydroxy-phenyl)-acetamide (**3a**) were synthesized. The identification of derivatives 4-hydroxy-chromen-2-one (**1a**, **2a**, **3a**), is made by using melting point, IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR spectra and elemental analysis. Melting point was determined on an electrothermal apparatus (Fisher Scientific 2555) in an open capillary tube and are uncorrected. Infrared spectra were recorded in cm<sup>-1</sup> for KBr pellets on a FT-IR Shimadzu 8400S spectrophotometer with resolution 4 cm<sup>-1</sup>. <sup>1</sup>H NMR spectra were recorded on a Bruker UNITY plus-500 'NMR 1' spectrometer using DMSO-d<sub>6</sub> as the solvent and TMS as the internal references standard ( $\sigma = 0, 00$  ppm). Chemical shifts are expressed in  $\delta$  ppm. Mass spectra were taken on a LKB 9000 mass spectrometer. Elemental analysis was performed on a Perkin-Elmer 240 BCHN analyzer. The purity of the compounds (synthesized) was routinely checked by TLC using Merck

Kieselgel-60 (F-254) and benzene, toluene, glacial acetic acid (80:10:10) as mobile phase. The spots were exposed in iodine vapor for visualization.

### Preparation of 4-hydroxy-3-(3-Hydroxy-benzoyl)-chromen-2-one (1a)

For this synthesis is used as substrate 4-hydroxy-chromen-2-one in a 100 ml flask mixed 3g 4-hydroxy-chromen-2-one, 2 g aldehyde salicylic, 8ml C<sub>2</sub>H<sub>5</sub>OH. The mixture was refluxed at 110 °C for ca. 5h. The obtained crystals yellow are filtered and rinsed with ethanol and dried at room temperature. Recrystallization form absolute ethanol gave a yellow product of 80% yield, melting point 253 °C.



Scheme-1

### Preparation of 3-(3-Hydroxy-benzoyl)-4-(2-hydroxy-phenylamino)-chromen-2-one (2a)

In a 100 ml flask were mixed 3g 3-(3-Hydroxy –benzoyl)-chromen-2-one with 10ml C<sub>2</sub>H<sub>5</sub>OH, 3g amino phenol .The mixture was refluxed at 100 °C for ca. 4h . The obtained red crystals are filtered and dried at room temperature. Recrystallization form C<sub>2</sub>H<sub>5</sub>OH gave red crystals product of 70 % yield, melting point, 373 °C.

### Preparation of N-[3-(3-Hydroxy-benzoyl)-2-oxo-7-thioureido-2H-chromen-4-yl]-N-(2-hydroxy-phenyl)-acetamide (3a)

In a 100 ml flask were mixed 2g of 3-(3-Hydroxy-benzoyl)-4-(2-hydroxy-phenylamino)-chromen-2-one ,with 8 ml C<sub>2</sub>H<sub>5</sub>OH , 1g tioure , 2ml SOCl<sub>2</sub> as catalyzer. The mixture was refluxed at 80 °C in water bath for ca. 6 h .The flask was placed in an ice bath for 1h until yellow crystalline precipitate was formed. After filtration the product was recrystallized from CH<sub>3</sub>CN . The recrystallization gave a red product at 70% yield, melting point; 180 °C.

## RESULTS AND DISCUSSION

By reacting equimolar amounts of 4-hydroxy-chromen-2-one and corresponding reagents (according Scheme-1) under reflux reaction conditions product 1a is synthesized in 80 % yield.

By reacting equimolar amounts of 4-hydroxy-3-(3-Hydroxy-benzoyl)-chromen-2-one and corresponding reagents (Scheme-2) under reflux reaction conditions product 2a is synthesized in 70 % yield.

By reacting equimolar amounts of 3-(3-Hydroxy-benzoyl)-4-(2-hydroxy-phenylamino)-chromen-2-one and corresponding reagents (Scheme-3) under reflux reaction conditions product 3a is synthesized in 80% yield.

The structure of 4-hydroxy-chromen-2-one derivatives (**1a**, **2a** and **3a**) were determined from their IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR spectra and their melting points as follows-

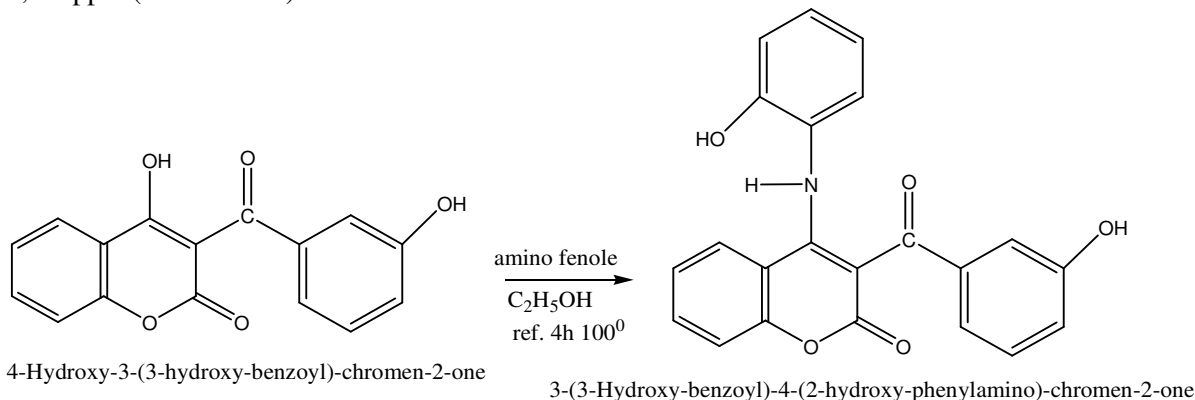
**1a:** IR bands (KBr, cm<sup>-1</sup>) 3850-2400cm<sup>-1</sup> (OH), (C-H stretch.), 1720 cm<sup>-1</sup> (C=O), 1600 (C=C stretch.), 750 cm<sup>-1</sup> (C-H bend.) 600 cm<sup>-1</sup> (C-Cl stretch.)

<sup>1</sup>H NMR (DMSO-d<sub>6</sub>) δppm; 9.68 ppm s (H, CHO), 7.21-7.53 t (H, aromatic), 5.18 s (H, OH)

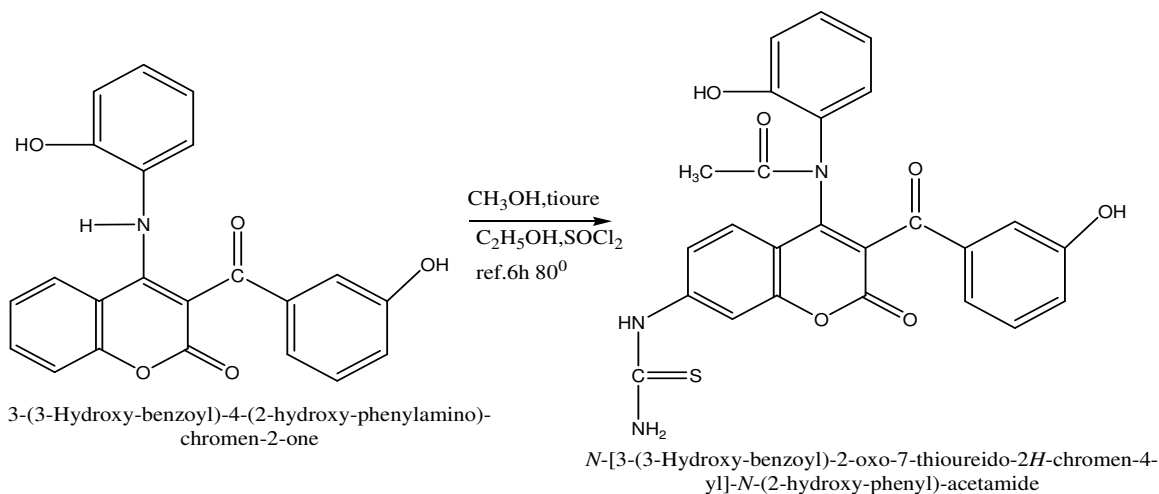
<sup>13</sup>C NMR (DMSO) δppm ; 166.9ppm (C-Cl), 162ppm (C,COO) ; 152ppm (C,C-O) ; 133.4 (C,C-Cl); 121.7, 125.6 , 128.0 (3C-aromatic)

**2a:** IR bands (KBr,  $\text{cm}^{-1}$ )  $3400\text{cm}^{-1}$  (OH)  $3200\text{cm}^{-1}$  (N-H stretch.),  $3000\text{cm}^{-1}$  (C-H stretch.),  $3200\text{cm}^{-1}$  (N-H stretch.),  $2730\text{cm}^{-1}$  (C-H stretch.),  $1725\text{cm}^{-1}$  (C=O stretch.),  $1600\text{cm}^{-1}$  (C=C stretch.),  $1050\text{cm}^{-1}$  (C-O stretch),  $750\text{cm}^{-1}$  (C-H bend.)

$^1\text{H}$  NMR (DMSO- $d_6$ )  $\delta$ ppm 6.37, 6.39, 7.41 t (3H aromatic) 5.0 (H, OH), 4.0 d (H, NH), 4.0 ppm s (NH)  
 $^{13}\text{C}$  NMR (DMSO)  $\delta$ ppm 181 ppm (C, C-NH), 162 ppm (C, COO), 151 ppm (C, C-O), 105, 109, 116, 127 ppm (4C aromatic).



Scheme-2



Scheme-3

Table-1 Analytical data

Compd.	m.p	M.F	% Elemental analysis				
			Found (Calculated)				
			C	H	N	O	S
1a	253 °C	C <sub>15</sub> H <sub>11</sub> NO <sub>3</sub>	71.14 (71.00)	4.38 (4.30)	5.53 (5.50)	18.95 (18.90)	
2a	373 °C	C <sub>22</sub> H <sub>15</sub> NO <sub>5</sub>	70.77 (70.71)	4.05 (4.00)	3.75 (3.70)	21.43 (21.40)	
3a	489 °C	C <sub>25</sub> H <sub>19</sub> N <sub>3</sub> O <sub>6</sub>	61.34 (61.29)	3.91 (3.90)	3.75 (3.70)	19.81 (19.75)	6.55 (6.50)

**3a:** IR bands (KBr,  $\text{cm}^{-1}$ ) 3280  $\text{cm}^{-1}$  (O-H stretch.), 3180  $\text{cm}^{-1}$  (NH stretch.), 3000  $\text{cm}^{-1}$  (C-H stretch.), 2400  $\text{cm}^{-1}$  (O-H carbocyclic), 1760  $\text{cm}^{-1}$  (C=O stretch.), 1600  $\text{cm}^{-1}$  (C=C stretch.), 1710  $\text{cm}^{-1}$  (C=O), 1020  $\text{cm}^{-1}$  (C-O), 750  $\text{cm}^{-1}$  (C-H bend.)

$^1\text{H}$  NMR (DMSO- $d_6$ )  $\delta$ ppm 7.4, 6.5, 6.4 (3H aromatic), 5.0 (H.OH), 4.0 s(H,NH), 3.53ppm t(2CH<sub>2</sub>OH), 2.65ppm t(3H,CH<sub>3</sub>N), 1.48-1.52ppm t(3H,2CH<sub>3</sub>), 1.40-1.55 ppm t(4H,2CH<sub>2</sub>)

$^{13}\text{C}$  NMR (DMSO)  $\delta$ ppm 176.0ppm (C,COOH), 167.ppm (C,C-NH), 162.0 (C,C=O), 151.7ppm(C,C-O), 127,109,105ppm (3C aromatic), 51.6(C,C-N), 46.6(C,C-N), 62.7(C,C-OH), 30.6,27.8ppm (C,CH<sub>3</sub>)

Table-2: Zone of inhibition (mm) of Antibacterial activity in *Staphylococcus aureus*  
Inhibition zone (mm)

Compound	2mg/ml	3mg /ml	5mg/ml
1a	12	16	18
2a	13	17	20
3a	14	17	21
Cephalexin	8	8	8
Streptomycin	20	20	20

Table-3: Zone of inhibition (mm) of Antibacterial activity in *E.Coli*  
Inhibition zone (mm)

Compound	2mg/ml	3mg /ml	5mg/ml
1a	8	13	17
2a	9	15	19
3a	10	19	18
Cephalexin	8	8	8
Streptomycine	20	20	20

Table-4: Zone of inhibition (mm) of Antibacterial activity in *Bacillus cereus*  
Inhibition zone (mm)

Compound	2mg/ml	3mg /ml	5mg/ml
1a	8	13	18
2a	9	14	19
3a	12	18	20
Cephalexin	9	9	9
Streptomycine	20	20	20

### Antibacterial activity

The purified synthesized compounds (**1a**, **2a**, **3a**) were subjected to test in vitro its antibacterial activity against three bacterial cultures; *Staphylococcus aureus*, *E.Coli* and *B.cereus*. Antibacterial activity of compounds was investigated applying the Kirby-Bayer method or disc method (d=5.5 mm max. capacity 10  $\mu\text{g}$ ).

### CONCLUSION

From the results the following conclusions were drawn:

- The study provides the first evidence that compounds (**1a**, **2a**, **3a**) obviously inhibit the growth of *S.aureus*, *E.coli* and *B.cereus*.
- The compounds (**1a**, **2a**, **3a**) compared with the antibacterial activity of Streptomycin in *S.aureus*, *E.coli* and *B.cereus*.
- This study provided the first evidence that these compounds **1a**, **2a**, **3a** showed a significant antibacterial effect against *S.aureus*, *E.coli* and *B.Cereus*.

- The chemical structures of synthesized compounds were determined according to extensive NMR experiments and published data.

### ACKNOWLEDGEMENT

The authors thank Prof. Branko Stanovnik, University of Ljubljana and its laboratory staff for <sup>1</sup>H NMR spectrum and elemental analyses.

### REFERENCES

1. S.Govori, V.Kalaj, V.Rapiq, L.Kalaj, and S. Dakovic, *Heterocycl. Commun.*, **8**, 129 (2002).
2. B.Stanovnik, H. Susachitzky, F.Scriven, *Progress in Heterocyclic Chemistry, Pergamon Press, Oxford*, **5**, 75(1993).
3. S.Lee, S.Shin, S.Kim and S.Kan, *Arch. Pharm. Res.*, **7**, 26(2003).
4. K.Vyas, K.Nimavat, R.Jani, and V.Hathi, *Orbital.*, **1**, 183(2009).
5. V.Abyshev, A.Gimdein, V.Semenov, E.M.Ageve, A. Abdulla, *Pharm. Chem. J.*, **40**, 607 (2006).
6. D.Aytemir, C. Hider, D. Erol, M. Ozalp and M. Ekizoglu, *Turk. J. Chem.*, **27**, 445(2003).
7. M.El.Saghier, B.Naili, K.Rammash, H. Saleh and M.Kreddan., *Arkivoc.*, **9**, 83(2012).
8. Z. M..Nofal, M.El-Zaha and Abd. El Karim M., *Molecules.*, **5**, 99(2000).
9. K.C.Chaluvaraju and K. Ishwarbhat Asian., *J. Chem.*, **20**, 4335(2008).
10. R. Ra Kali, S. Grworamma and B. Suresh, *Asian J. Chem.*, **20**, 5289(2008).
11. A. Mohammed Ashraf and Sh.Mohammed, *Boorg. Med. Chem. Lett.*, **17**, 3314(2009).
12. Z.M.Nofal, M.El-Zahar, S. Abd El-Karim, *Molecules.*, **5**, 99(2000).
13. K. B. Vyas, K. S. Nimavat, G. R. Jani, M. V.
14. Hathi, *Orbital.*, **1**, 183(2009).

[RJC-1458/2016]