

## HYDROCHALCONE COMPOUNDS FROM INDONESIAN MEDICINAL PLANT, “SIRIH HUTAN”, *Piper aduncum* (*Piperaceae*)

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### ABSTRACT

Isolation, purification and identification of two hydrochalcone compounds from Indonesian medicinal plant “Sirihhutan”, *Piper aduncum* (*Piperaceae*) had been done. Isolation and purification of ethylacetate extract subjected to column chromatography (SiO<sub>2</sub>; (i).n-hexane-ethylacetate = 10:1 ~ 1 : 1; ethylacetate (ii).n-hexane-ethylacetate = 8: 1 gave two pure compounds. Based on infra red, 1D & 2D-NMR, mass spectral data and comparison chemical shift of protons and carbons, the isolated compounds are 2',6'-dihydroxy-4'-methoxy dihydrochalcone and 2',6',4-trihydroxy-4'-methoxy dihydrochalcone with free radical scavenging inhibition values are 21.77 % and 90.1 % respectively.

**Keywords:** Hydrochalcone; *Piper aduncum*; *Piperaceae*; 2',6'-dihydroxy-4'-methoxy dihydrochalcone; 2',6',4-trihydroxy-4'-methoxy dihydrochalcone

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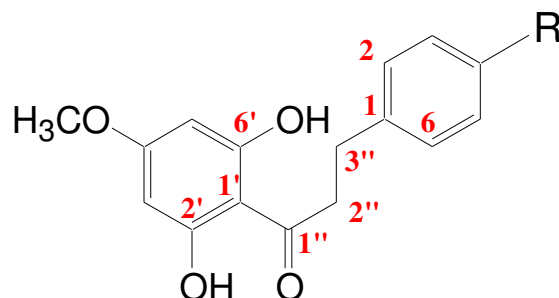
### INTRODUCTION

This research is a continuation of chemical studies on Indonesian medicinal plants, especially on bioactive chemical compounds as antioxidant.<sup>1-5</sup> This research focused on hydrochalcone compounds from “SirihHutan” (localname), *Piper aduncum* L. (*Piperaceae*). Traditionally, this plant is used as a deodorant, antihemorrhagea, antiemetic, anti-septic, antibacterial and antifungal.<sup>6</sup> Based on the literature, this plant has activities as antibacteria<sup>6</sup>, insecticide<sup>7</sup>, and antioxidant<sup>8</sup>.

Isolation results of antioxidant bioactive compounds from ethylacetate fraction of *P. aduncum* L based on free radicals scavenging activity test using DPPH (1,1-diphenylpicrylhydrazine) gave two hydrochalcone compounds, i.e. 2',6'-dihydroxy-4'-methoxy dihydrochalcone and 2',6',4-trihydroxy-4'-methoxy dihydrochalcone.



Fig.-1: Sirihhutan, *Piper aduncum* (*Piperaceae*)



R = H, 2',6'-dihydroxy-4'-methoxy dihydrochalcone (1)

R = OH, 2',6',4'-trihydroxy-4'-methoxy dihydrochalcone (Asebogenin) (2)

Fig.-2: Chemical Structures of the Isolated Compound from "Sirihutan" *Piper aduncum*

## EXPERIMENTAL

### Materials

The leaves of *Piper aduncum* were collected from Samarinda forestry, East Kalimantan, Indonesia. The spectra of IR were taken with Hitachi 260-30 and JASCO FT-IR-5300 spectrophotometers. The  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra were measured with Jeol GX-500 spectrometer (500 MHz), while mass spectra (MS) were measured with LCMS (Q Micro QQA 842 1.0). Column chromatography was carried out by using silica gel 60 (70-230 mesh ASTM, Merck) as absorbent. Thin layer chromatography (TLC) was conducted on precoated Kieselgel 60 GF<sub>254/366</sub> plate (0.2 mm, Merck). Spot on the TLC plates were detected by reagent spray 1%  $\text{Ce}(\text{SO}_4)_2/10\% \text{H}_2\text{SO}_4$  followed with heating at 110 °C.

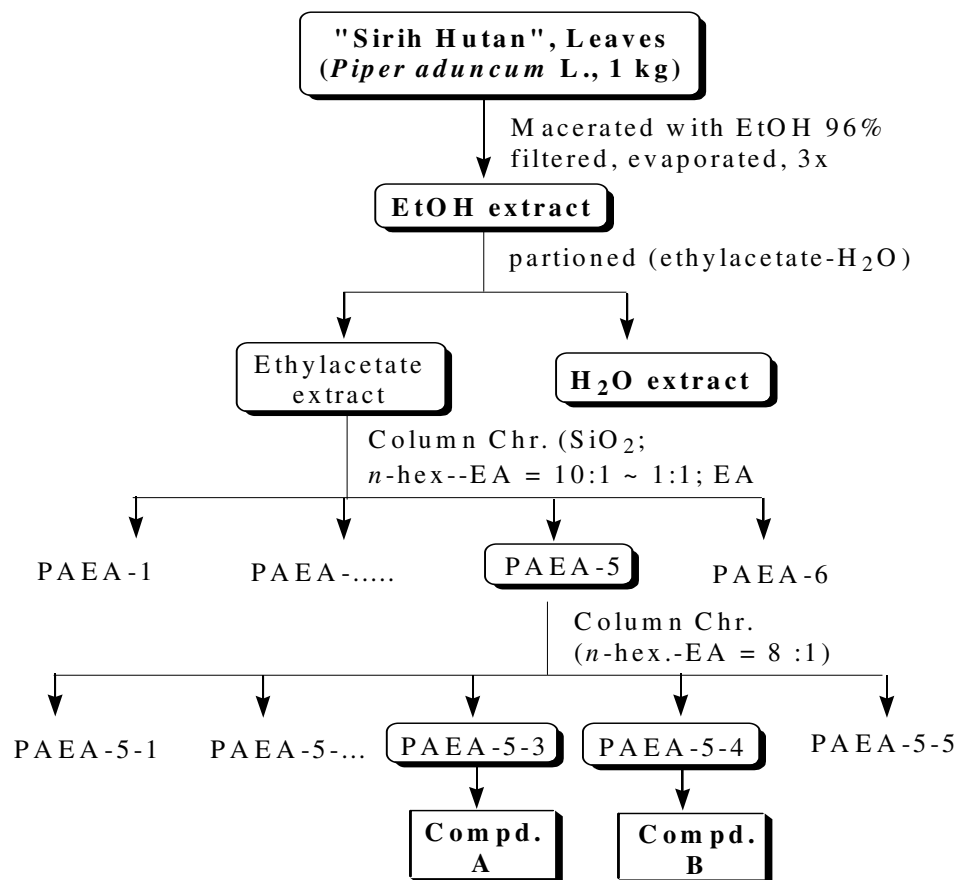


Fig.-3: Isolation Procedure for Hydrochalcone Compounds

### Extraction, Isolation and Purification

The air-dried leaves (1 kg) of *P. aduncum* collected from Samarinda Forestry, East Kalimantan, Indonesia were extracted with ethanol 96% by maceration. The solvents were evaporated under reduced pressure to give ethanol extract (78 g; 7.8%). The ethanol extracts were partitioned with ethylacetate-water = 1 : 1 and evaporated to give water extract (45.1g; 4.51%) and ethylacetate extract (30.8 g; 3.07%). The ethylacetate extracts (10 g) were then subjected to column chromatography [ $\text{SiO}_2$ ; n-hexane-ethylacetate = 10 : 1 ~ 1 : 1, ethylacetate and ethanol successively] to provide fractions. 1 (PAEA-1) ~ 6 (PAEA-6).

Fraction PAEA-5 was separated by column chromatography [ $\text{SiO}_2$ , n-hexane-ethylacetate = 5 : 1] to give compound A (20,2 mg), B (35,1 mg) and other fractions.

### RESULTS AND DISCUSSION

Compound A (**1**) was obtained in white powder form and has a molecular ion peak at  $m/z$  272 ( $\text{M}^+$ ) in its mass spectrum by LC-MS. Infra Red (IR) spectrum of compound A showed absorption band at  $1650\text{ cm}^{-1}$  (carbonyl group) and  $3265\text{ cm}^{-1}$  (hydroxyl group). The  $^1\text{H-NMR}$  spectrum of compound **1** showed signals due to one methoxy group at  $\delta\text{H}$  3.78 (s); two methylene groups at  $\delta\text{H}$  3.02 (t,  $J=8.5$ ; 7.0 Hz);  $\delta\text{H}$  3.79 (d,  $J=8.5$ ; 7.0 Hz). Furthermore, the chemical shift of proton on a benzene ring was observed at  $\delta\text{H}$  5.93 (m);  $\delta\text{H}$  7.19 (m);  $\delta\text{H}$  7.26 (m). Distortionless enhancement by polarization transfer (DEPT) experiment on compound **1** by  $^{13}\text{C-NMR}$  spectroscopy disclosed the presence of one methoxy carbon, two methylene carbons, seven methane carbons, six quaternary carbons and a carbonyl carbon (Table-1). The  $^{13}\text{C-}^1\text{H}$  COSY experiment revealed connectivity between respective protons and carbons. The plane structures of compound **1** were constructed by H-H COSY experiment of **1** estimated the presence of correlation between H-2, H-3, H-4, H-5 to H-6 and between H-2'' to H-3''. In the HMBC experiment, compound **1** was shown to have correlations between H-2'' to C-1'', C-3'', C-1; H-3'' to C-2'', C-1'', C-1, C-2, C-3; H-5' to C-6', C-1', C-4'; H-3' to C-2', C-1' (Fig.-4).

Based on IR, NMR and mass spectral data and supported by comparison data on the chemical shift of protons and carbons by Masuoka *et al.*, (1997), compound A (**1**) was determined as 2',6'-dihydroxy-4'-methoxy dihydrochalcone.

Compound B (**2**) was an amorphous powder form. It gave molecular ion peak at  $m/z$  288 with molecular formula  $\text{C}_{22}\text{H}_{32}\text{O}_5$  indicate the addition of hydroxyl groups to compound **1**. The IR spectrum of **2** showed absorption bands due to the hydroxyl group ( $3320\text{ cm}^{-1}$ ),  $1672\text{ cm}^{-1}$  and  $1561\text{ cm}^{-1}$  due to C=O and C=C stretching. The proton and carbon NMR spectra showed no significant chemical shift difference between compound **1** and compound **2**, unless there is a chemical shift in the C-4 of  $\delta\text{C}$  126,10 (d) on compound **1** to be  $\delta\text{C}$  157.8 (s) on compound **2**.

Table-1: Chemical Shift ( $\delta_{\text{H}}$ ) of Compound **1** and **2** compared to 2',6'-dihydroxy-4'-methoxy dihydrochalcone<sup>8</sup>

No	Isolated Compound (1) J in Hz	2',6'-dihydroxy-4'-methoxy dihydro chalcone <sup>8</sup>	Isolated Compound (2)	2',6',4'-trihydroxy-4'-methoxy dihydrochalcone <sup>8</sup>
1	-	-	-	-
2	7.27 (m)	7.23 (m)	6.71 (d, $J=2.0$ )	6.40 (d, $J=2.0$ )
3	7.27 (m)	7.23 (m)	7.11 (d, $J=2.0$ )	7.04 (d, $J=2.0$ )
4	7.18 (t, $J=2.21$ )	7.16 (t-like)	-	-
5	7.26 (m)	7.23 (m)	7.21 (d, $J=2.0$ ; 6.4)	7.04
6	7.19 (m)	7.23 (m)	6.00 ( $J=2.0$ )	6.40
1'	-	-	-	-
2'	-	-	-	-
3'	6.00 (s)	5.92 ( $J=13.2$ )	5.89 (s)	5.92 (s)
4'	-	-	-	-
5'	5.93 (s)	5.92 ( $J=13.2$ )	5.71 (s)	5.92 (s)
6'	-	-	-	-
1''	-	-	-	-
2''	3.39 (t, $J=7.75$ ; 7.80)	3.33 (m)	3.29 (m)	3.27 (m)

3''	3.02 (t, J=8.45; 7.1)	2.94 (m)	2.89 (m)	2.85 (m)
4'- OMe	3.78 (s)	3.75 (s)	3.79 (s)	3.75 (s)

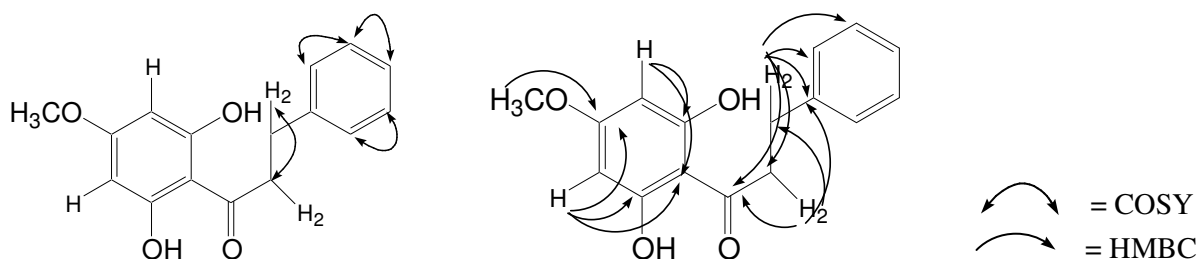


Fig.-4: H-H COSY Analysis (left) and HMBC Analysis (right) for Compound 1

Tabel-2: Chemical Shift  $\delta\text{C}^{13}\text{C-NMR}$  for Compound 1 and 2

CarbonNumber	Isolated Compound (1)	Isolated Compound(2)	2',6',4-trihydroxy-4'-methoxy dihydrochalcone
1	141.79 (s)	134.7 (s)	135.1 (s)
2	128.56 (d)	131.1 (d)	132.2 (d)
3	129.40 (d)	116.4 (d)	116.6 (d)
4	126.10 (d)	157.2 (s)	157.8 (s)
5	129.40 (d)	116.4 (d)	116.6 (d)
6	128.56 (d)	131.1 (d)	132.2 (d)
1'	104.95 (s)	106.8 (s)	108.1 (s)
2'	165.71 (s)	166.3 (s)	167.8 (s)
3'	94.48 (d)	95.1 (d)	95.9 (d)
4'	165.71 (s)	168.2 (s)	169.1 (s)
5'	94.48 (d)	95.1 (d)	95.7 (d)
6'	165.71 (s)	166.3 (s)	168.2 (s)
1''	204.74 (s)	207.5 (s)	207.8 (s)
2''	45.76 (t)	48.2 (t)	48.4 (t)
3''	30.69 (t)	32.1 (t)	33.1 (t)
4'-OMe	55.65 (q)	56.6 (q)	56.8 (q)

### Free Radical Scavenging Activity

The result of antioxidant activity using free radical scavenging activity method for all extracts (ethanol, ethylacetate and water), some fractions and pure compounds (**1**, **2**) obtained from column chromatography analysis can be seen in Table-3. Compound **1** (2',6'-dihydroxy-4'-methoxy dihydrochalcone) showed no antioxidant activity, while Compound **2** (2',6',4-trihydroxy-4'-methoxy dihydrochalcone) showed activity at 90.1 %.<sup>8</sup>

Table-3: Free Radical Scavenging Activities of all Extracts and Fractions

No.	Sample Name	Inhibition (%)	Level <sup>14*</sup>
1	Ethanol extract	88.65	Active
2	Fr. Ethylacetate	89.42	Active
3	Fr. Water	96.73	Active
4	Fr. PAEA-1	7.24	No
5	Fr. PAEA-2	21.33	No
6	Fr. PAEA-3	34.0	No
7	Fr. PAEA-4	83.3	Active
8	Fr. PAEA-5	84.31	Active

9	Fr. PAEA-6	72.03	Active
10	Fr. PAEA-5-1	34.12	No
11	Fr. PAEA-5-2	27.35	No
12	Fr. PAEA-5-3 (Compound1)	21.77	No
13	Fr. PAEA-5-4 (Compound2)	90.1	Active
14	Fr. PAEA-5-5	17.56	No

\*) :0 - 50 %= no active;

>50 % = active

### CONCLUSION

Two hydrochalcone compounds from Indonesian medicinal plant “Sirih Hutan”, *Piper aduncum* have been isolated and identified as 2',6'-dihydroxy-4'-methoxy dihydrochalcone and 2',6',4-trihydroxy-4'-methoxydihydrochalcone. One of them is 2',6',4-trihydroxy-4'-methoxy dihydrochalcone has free radical scavenging inhibition of 90.1 %, potentially as an antioxidant compound.

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