

# SYNTHESIS OF 3-(3-METHOXY-2-NITRO-3-PHENYLPROPYL)-9H-CARBAZOLE FROM FRIEDEL-CRAFTS REACTION

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#### ABSTRACT

A new method for the Synthesis of 3-(3-methoxy-2-nitro-3-phenylpropyl)-9H-carbazole *via*Friedel-Crafts reaction through nitroolefins derivatives. This the first type Friedel-Crafts reaction of the nitroolefin derived from Bayliss-Hillman mediated by con.  $H_2SO_4$  thus giving a simple synthesis of substituted olefin derivatives. Although various electron deficient alkenes such as  $\alpha,\beta$ -unsaturated carbonyl compounds, nitriles, sulfones and phosphonates have been employed as substrates in the nitroolefin reaction. This technology also opens new opportunity to make application of compounds.

**Keywords:** Friedel-Crafts reaction, nitroolefin, carbazole, Methanol, K<sub>2</sub>CO<sub>3</sub>, nitro methane, methane sulphonic acid, benzaldehyde.

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# **INTRODUCTION**

The reaction is a well-known coupling reaction of aldehydes and activated olefins catalyzed by tertiary amines or tertiary phosphines.<sup>1-5</sup> The B-H reaction is a new carbon-carbon bond forming reaction producing a functional class of well-applied molecules and the adducts have been employ for various organic transformations.<sup>6-10</sup> The Baylis-Hillman adducts have been successfully utilized as synthons in numerous named reactions for example Heck reaction, Diels-Alder reaction, Aldol condensation, Claisen rearrangement, Friedel-Crafts reation<sup>11-15</sup> etc. The Friedel-Crafts reaction is one of the most commonly used reactions in organic chemistry whose applications in intellectual as well as industrial fields have been well documented.<sup>16, 17</sup> The Friedel-Crafts reaction on Baylis-Hillman reaction was to begin with reported by Basavaiah et al, following his pioneering work different reports have appeared in association with Friedel-Crafts chemistry on Baylis-Hillman reaction. The nitroaldol or Henry reaction is a standard aldol-type reaction connecting an aldehyde and a nitroolefin.<sup>18-20</sup>

# EXPERIMENTAL

# 3-(3-methoxy-2-nitro-3-phenylpropyl)-9H-carbazole

To a stimulated solution of (*E*)-3-(2-nitro-3-phenylallyl)-9H-carbazole (**1**) (0.52g, 2 mmol) in MeOH (10 mL) and K<sub>2</sub>CO<sub>3</sub> (0.28g, 2 mmol) was added at RT. The reaction mixture was motivated at RT for 2 h. After achievement of reaction, the mixture was moved into the water and the aqueous layer was extracted with ethyl acetate ( $3 \times 10$  mL). The collective organic layer was washed with brine (20 mL), and dried in excess of anhydrous Na<sub>2</sub>SO<sub>4</sub> and the combined organic layer was evaporated. The plain product thus obtained was purified by column chromatography (5%, EtOAc/hexanes) to supply the preferred compound 3-(3-methoxy-2-nitro-3-phenylpropyl)-9H-carbazole (**3a**) in excellent yield (67%) as a colorless solid.

**3-(3-methoxy-2-nitro-3-phenylpropyl)-9H-carbazole (3a)** Yield: 65 %, IR (KBr): 3458, 1635, 1572, 1526. cm<sup>-11</sup>, HNMR: δ 2.65 (d, 2H), 2.97 (s, 3H),3.27 (dd, 1H), 4.36 (d, 1H), 7.31-7.92 (m, 12H), 10.34

*Rasayan J. Chem.*, 11(1), 321-323(2018) http://dx.doi.org/10.7324/RJC.2018.1111759 (bs, 1H). <sup>13</sup>CNMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  17.24, 19.31, 30.77, 57.16, 125.85, 126.92, 130.23, 120.31, 130.84, 131.42, 131.96, 132.25, 133.15, 134.62, 135.23, 135.55.Mass (m/z): 360 (M<sup>+</sup>) Anal. Calcd for C<sub>22</sub>H<sub>20</sub>N<sub>2</sub>O<sub>3</sub>C, 73.32; H, 5.59; N, 7.77Found C, 73.35; H, 5.60; N, 7.75.

#### **RESULTS AND DISCUSSION**

To implement our idea, first we have selected the Baylis-Hillman adduct (*E*)-2-nitro-3-phenylprop-2-en-1-ol (1) consequent from benzaldehyde and nitro methane as starting material for the Friedel-Crafts reaction. The greatest results were obtained when the addition of a catalytic amount of methane sulphonic acid to the solution of a B.H adduct in 9H-carbazole at room temperature, productively led to the preferred product (*E*)-3-(2-nitro-3-phenylallyl)-9H-carbazole The fresh carbazole product was treated with methanol and K<sub>2</sub>CO<sub>3</sub>at room temperature led to the beloved anticipated 3-(3-methoxy-2-nitro-3phenylpropyl)-9H-carbazole**3a** in excellent yield (Scheme-1).



The synthesized carbazole creation **3a** was characterized by <sup>1</sup>H &<sup>13</sup>C NMR spectral analyses. The <sup>1</sup>H NMR spectrum of compound **3a** showed one doublet for the aryl methyl protons at  $\delta$  2.65 and CH-NO<sub>2</sub> appear as a double doublet at  $\delta$  3.27. The methoxy proton was experimental as a singlet at  $\delta$  2.97 and CHOCH<sub>3</sub> acetylene protons came out as a doublet at  $\delta$  4.36. The aromatic protons appeared as multiplets in the section of  $\delta$  7.31-7.92 and the NH proton appeared as a broad singlet at  $\delta$  10.34.

Expectant by this results we organized a variety of carbazole and successfully transformed them into their consequent methoxy derivatives **3b-j**, according to Scheme-2.



 $\begin{array}{l} {\sf R} = \mbox{ 2-Me } 2\mbox{-MeO}, \mbox{ 4-MeO}, \mbox{ 3,4-(MeO)}_2 \\ {\sf 3,4-(OCH_2O)}, \ \mbox{ 2-Cl-, 3-Cl}, \mbox{ 4-Cl}, \mbox{ 3,4-(Cl)}_2 \end{array}$ 

Scheme-2

# CONCLUSION

In conclusion, this technology represents the Friedel-Crafts reaction of the Baylis-Hillman reaction derived from nitroolefins. A simple synthesis of 3-(3-methoxy-2-nitro-3-phenylpropyl)-9H-carbazole derivatives. Variety of carbazole was used for the first time in Nucleophilic replacement reaction on Baylis-Hillman adducts derived from nitro olefins. This technology also opens new opportunity to make a library of compounds.

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S.No.	Carbazole	Methyloxy	Yield
1a	C <sub>6</sub> H <sub>5</sub>	3a	67
1b	2-Me	3b	69
1c	4-Me	3c	70
1d	2-MeO	3d	60
1e	4-MeO	3e	68
1f	3,4-(MeO) <sub>2</sub>	3f	71
1g	3,4-(OCH <sub>2</sub> O)	3g	70
1h	2-Cl	3h	68
1i	3-C1	3i	67
1j	4-Cl	3ј	68

Table-1: Synthesis of 3-(3-methoxy-2-nitro-3-phenylpropyl)-9H-carbazole from friedel-crafts reaction

<sup>a</sup>All reactions were accepted with 2 mmol scale of carbazole (1a-j),<sup>b</sup>All products gave acceptable IR, <sup>1</sup>H NMR (300 MHz), <sup>13</sup>C NMR (75 MHz), mass spectral data and elemental analyses. <sup>c</sup>Yields of the clean products (3a-j) obtained following column chromatography (silica gel, (3a-j) 5% EtOAc in hexanes

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#### REFERENCES

- 1. D. Basavaiah, B.S. Reddy and B.S. Badsara, *Chemical Review*, **110**, 5447(2010).
- 2. V.F. Alejandro, G.Y. Cristina, A.V. Jesffls and A.E. Miguel Angewanda Chemical, 49, 4278(2010).
- 3. M.E. Krafft and T.F.N. Haxell, Journal of American Chemical Society, 127, 10168(2005).
- 4. E.L. Richards, P.J. Murphy, F. Dion, S. Fratucello, P.M. Brown, T. Gelbrich, and M.B. Hursthouse, *Tetrahedron*, **57**, 7771(2001).
- 5. M.C. Redondo, M. Ribagorda and M.C. Carreno, Organic. Letter, 12, 568 (2010).
- 6. B.G. Jellerichs, J.R. Kong and M.J. Krische, *Journal of American Chemical Society*, **125**, 7758 (2003).
- 7. D.Y. Park, S.J. Kim, T.H. Kim and J.N. Kim, *Tetrahedron Letter*, 47, 6315(2006).
- 8. G.P. Blacks, F. Dion, S. Fratucello, P.J. Murphy, M. Nielsen, H.L Williams and N.D. A. Walshe, *Tetrahedron Letter*, **38**, 8561(1997).
- 9. S. Gowrisankar, H.S. Lee, S.H. Kim, K.Y. Lee and J.N. Kim, Tetrahedron, 65, 8769(2009).
- 10. A. Faltin, C.; Fleming, E. M.; Connon, S. Journal of Organic Chemistry., 69, 6496(2004).
- 11. G.P. Blacks, F. Dion, S. Fratucello, P.J. Murphy, M. Nielsen, H.L. Williams and N. D.A. Walshe, *Tetrahedron Letter*, **38**, 8561(1997).
- 12. K. Pavan and D. V. Yashwant, Journal of Organic Chemistry, 75, 8457(2010).
- 13. P.R. Krishna, V. Kannan and G.V. Sharma, Journal of Organic Chemistry, 19, 6467(2004).
- 14. K. Bera, and N.N. IrishiNamboothiri., Asian Journal Organic Chemistry, 3, 1234(2014).
- 15. D. Basavaih, M. Krishnamacharyulu, R. SugunaHymaand S. Pandiyaraju, *Tetrahedron Letter*, **38**, 2141(1997).
- 16. B. Sunita and S. HarLal, Rasayan Journal Chemistry. 9, 614, (2016).
- 17. S. Monica, Charlena and I. H. Suparto, Rasayan Journal Chemistry, 9, 650, (2016).
- 18. G. M. Srirangam and K. Parameswara Rao, *Rasayan Journal Chemistry.*, **10**(1), 46 (2017)
- 19. K. Shakti Singh, G. K. Lohiya and V. K. Mourya, Rasayan Journal Chemistry, 2, 451 (2009)
- 20. K. Sivaram, M. C. Rao, G. Giridhar, M. Tejaswi, B. T. P. Madhav, V. G. K. M. Pisipati and R. K. N. R. Manepalli, *Rasayan Journal Chemistry*, **9(4)**, 697 (2017).

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