

# SYNTHESIS AND ANTIMICROBIAL ACTIVITY OF 3-(2-(4Z)-4-SUBSTITUTED BENZYLIDENE-4,5-DIHYDRO-5-OXO-2-PHENYL IMIDAZOL-1-YL)ETHYL)-6,8-DIBROMO SUBSTITUTED-2-SUBSTITUTED QUINAZOLINE-(3H)-ONE

R.Suthakaran\*, S.Kavimani<sup>1</sup> P.Venkaiiah<sup>2</sup> and K.Suganthi

\*Department of Pharmaceutical Chemistry, Vijaya college of Pharmacy-HYDERABAD-500511(A.P),INDIA

<sup>1</sup>-Department of Pharmacology, College of Pharmacy, Mother Teresa Institute of Health Sciences, PONDICHERY – 605 006,(PONDICHERY),INDIA

<sup>2</sup> - Department of Chemistry, School of Chemistry and Biotechnology, Shanmugha Arts, Science, Technology & Research Academy (SASTRA University), THANJAVUR- 613009, (T.N), INDIA

E.Mail: sudha\_sudhar@rediffmail.com

---

## ABSTRACT

6,8-Disubs.anthranilic acid ( $X=H, Br$ ) (**1**) reacted with acetic anhydride/ benzoyl chloride/ chloro acetyl choride ( $R_{a-f}$ ) to give 2-methyl / phenyl / chloro methyl disubstituted benzooxazine-4-one ( $X=H, Br$ ) derivatives respectively (**2**). Benzoyl glycine (**3**) reacts with substituted benzaldehyde (**4**) derivatives ( $R'_{1-5}$ ) to give 4 -substituted benzylidene-2-phenyl-oxazolone derivatives (**5**). These on reaction with ethylene diamine in presence of glacial acetic acid gave (**6**). The latter (**7**) have been synthesized by condensing (**2**) and (**6**), gave 30 imidazoloquinazoline-4- one derivatives. The structures of the compounds have been confirmed on the basis of their IR, <sup>1</sup>HNMR and MS spectral data. All the compounds (**7 a-ziv**) have been screened for their antimicrobial activities. Most of the compounds have shown promising antibacterial, and antifungal activity.

**Keywords:** Anthranilic acid, Benzooxazineone, quinazoline, Oxazolone, Imidazolone, Antibacterial, Antifungal.

---

## INTRODUCTION

A series of quinazoline analogs is designed to possess an alkyl, aryl and chloro alkyl functions at position 2, an alkyl amine (ethylene) bridge linked with Oxazole moiety it becomes Imidazole at position 3, a unsubstituted or dibromo substitution at 6 and 8 of Benzooxazine-4-one act as one pharmacophore. A series of Oxazolinone analogs, possess phenyl at position 2, H/ O-Cl/ O-OH/ P-(CH<sub>3</sub>)N/ P-OCH<sub>3</sub> substituted benzylidene at position 4 and an alkyl amine bridge (n=2) linked with Benzooxazine one moiety, it becomes quinazoline-4- one at position 1<sup>1-3</sup>. The 1-ethylamino function of Imidazole condensed with 2-methyl/phenyl/chloromethyl-6,8-un/ dibromo substituted Benzooxazine-4-one to give title compound. Methyl / phenyl/ chloromethyl substituents in quinazoline and hetero cyclic moieties like quinazoline and imidazole are known to contribute to the enhancement of the biological activities<sup>4-6</sup>.

The synthetic strategy to synthesize the targets **7(a-ziv)** around 30 compounds is depicted in scheme 1. In step 1, disubstituted anthranilic acid ( $X= H$  or  $Br$ ) (**1**) were allowed to react with

acetic anhydride / benzoyl chloride in dry pyridine / chloro acetyl chloride in dry benzene and followed by AC<sub>2</sub>O in few drops of pyridine for cyclization to produce 6,8- disubstituted (X= H or Br ) -2-methyl/phenyl/chloromethyl-4H-benzo [d] [1,3] oxazine-4-one ([X=H]) a, b, c, and ([X=Br]) d,e and f by using standard procedure. In step 2, benzoyl glycine were prepared from benzoyl Chloride and glycine in presence sodium hydroxide and followed by acid neutralization to get benzoyl glycine (3), The benzoyl glycine was subjected to condensed with phenyl/ Dimethyl amino phenyl / O- chloro phenyl/ O-hydroxy phenyl / P-methoxy phenyl aldehyde's (4) (R<sup>1-5</sup>) in presence of acetic anhydrite and sodium acetate to produce respective (4 E)-4- substituted benzylidene-2- phenyl oxazol-5(4H)- one (5 ). The oxazole derivative (5) was then reacted with ethylene diamine in presence of glacial CH<sub>3</sub>COOH to afford the (4E)-1-(2-amino ethyl)-4- benzylidene 2-phenyl-1h- imidazol-5(4H) - one (6) <sup>7-9</sup>.

In step3, the 6,8- disubstituted-2- methyl/ phenyl chloro methyl- 4H- benzo[d] [1,3] oxazin -4- one (2) were reacted with (4E) -1-(2- amino ethyl) – 4- substituted benzylidene-2-phenyl- 1H – Imidazole – 5 (4H)- one(6) to yield the title compounds ,structure elucidation of the synthesized intermediates and final products (7) was attained by the aid of IR, <sup>1</sup>H-NMR and MS spectrometry.

### EXPERIMENTAL

The melting points were determined on a MEL-Temp apparatus by open capillary method and are uncorrected. IR spectra (KBr disc) were recorded on Perkin –Elmer 1800 FT-IR spectrophotometer. <sup>1</sup>H-NMR spectra (CDCl<sub>3</sub>) were recorded on Perkin- Elmer R-32 (90 MHz) using TMS as internal standard. Mass spectra were measured with a FINIGAN TSQ-70 spectrometer at 70 e.V. purity of the compounds was checked by TLC using silica gel-G plates ,benzene- methanol and benzene- ethyl acetate as developing solvent.

#### Synthesis of 2-methyl/phenyl/chloromethyl – 4H – Benzo[d] [1,3] oxazin – 4 – one (2)

**R<sub>(a-f)</sub>**: A mixture of disubstituted anthranilic acid (X=H,Br) (1) (0.12 mol) in acetic anhydride in few drops of pyridine (0.2 mol)/ benzoyl Chloride in dry pyridine (0.3 mol)/Chloro acetyl Chloride in drug benzene (0.12 mol) and followed by acetic anhydride (0.3 mol) in pyridine was reflux for 3 hours. The excess solvents were then distilled off under reduced pressure. The reaction mixture was filtered, washed, dried and re-crystallized with absolute ethanol for methyl and phenyl compounds, where as chloro methyl compounds used a mixture of Chloroform and ethyl acetate.

#### Synthesis of (4Z) – 4 – substituted benzylidene – 2 – phenyl oxazol – 5 (4H) - one (5) R<sup>1-5</sup>:

Prepared crystalline benzoyl glycine (3) from standard procedures. Place a mixture of redistilled H/ O-Cl/ O-OH/ P-(CH<sub>3</sub>)N/ P-OCH<sub>3</sub> substituted benzaldehyde (0.25 mol) (4) with benzoyl glycine (0.25 mol) in acetic anhydride (0.75 mol) and anhydrous Sodium acetate was allowed to heat with constant shaking. As soon as the mixture has liquefied completely, transfer the flask to a water bath and heat for 2 hours. Then add 100 ml of ethanol slowly to the contents to stand for overnight,

wash with ice cold alcohol and then with boiling water, filtered and re-crystallized from Benzene.(5a) , , m.p. 482<sup>0</sup>k, Yield (85%), M+1 249, IR CM<sup>-1</sup>-3210,1550 lactone, and 1650 <sup>1</sup>H-NMR 7.14 – 7.30 C<sub>6</sub>H<sub>5</sub> , 7.3 –7.6 C<sub>6</sub>H<sub>5</sub> , 7.6 C<sub>6</sub>H<sub>5</sub>-CH<sub>2</sub>=.

**Synthesis of (4Z) – 1 – (2-amino ethyl) – 4 – Substituted benzylidene – 2 – phenyl -1H-imidazol – 5(4 H)– one (6) R<sup>1</sup>(1-5)**: A mixture of (Z) – 2 – Substituted benzylidene – 2 – phenyl oxazolidin – 5- one (0.1 mol) (5) in ethylene diamine (0.1 mol)(6) in presence of glacial acetic acid under anhydrous conditions reflux for 8 hours. The reaction mixture was filtered, dried and re-crystallized from absolute alcohol. (6a), m.p. 642<sup>0</sup>k, Yield (79%), M+1 291, <sup>1</sup>H-NMR

3.26, 3.46, 0.9, 7.3-7.6, IR  $\text{CM}^{-1}$ -3450, 2900, 1510, and 1650,  $^1\text{H-NMR}$  protons 7.14 – 7.30  $\text{C}_6\text{H}_5$ , 7.3-7.6  $\text{C}_6\text{H}_5\text{-CH}_2=$ , 3.22  $\text{CH}_2$ , 2.91  $\text{CH}_2$ , 2.0  $\text{NH}_2$ , 7.6  $\text{C}_6\text{H}_5\text{-CH}_2=$ .

**Synthesis of 3-(2-(4Z)-4-substituted benzylidene-4,5-dihydro-5-oxo-2-phenyl imidazol-1-yl)ethyl)-6,8-dibromo substituted-2-substituted quinazoline-(3H)-one (7)<sup>10</sup>:** The title compounds were prepared from (2) (0.1 mol) and (6) (0.1 mol) in presence of glacial acetic acid was refluxed under anhydrous conditions for 8 hours. The reaction mixture was cooled to room temperature, filtered, dried and re-crystallized from absolute alcohol. (7a), m.p. 907<sup>0</sup>k, Yield (67%), M+1 434,  $^1\text{H-NMR}$  3.26, 3.46, 0.9, 7.3-7.6, IR  $\text{CM}^{-1}$  3200, 1450, and 1650. The results of Physical and Spectral Data shown in Table-1, and Table-2 respectively.

## PHARMACOLOGICAL EVALUATION ANTIMICROBIAL ACTIVITY<sup>11-13</sup>

All the synthesized compounds were screened for their antibacterial and antifungal activities. For preliminary screening, the antimicrobial tests were carried out by the disc-diffusion method. Using Mueller-Hinton agar (MHA) medium and Sabouraud's dextrose agar (SDA) medium, for bacteria and fungi respectively. The discs (6mm in diameter), impregnated with the test compounds (25  $\mu\text{g/ml/disc}$  for bacteria and 1000  $\mu\text{g/ml/disc}$  for fungi). Negative controls were prepared using the same solvent (DMSO) employed to dissolve the test compounds. Ofloxacin (25  $\mu\text{mg/ml/disc}$ ) and Clotrimazole (1000  $\mu\text{g/disc}$ ) were used as positive reference standards to determine the sensitivity of each microbial species tested. The inoculated plates were incubated at 37<sup>o</sup> C for 24 h and 27<sup>o</sup> C for 72 h for bacterial and fungal strains respectively. Antimicrobial activity was evaluated by measuring the diameter of zone of inhibition against test organisms.

Based on the results (Table No.3), it is referred that quinazolone derivatives and have significant inhibition effect on the growth of bacteria like *Escherichia coli* (EC), *Pseudomonas aeruginosa* (PA) and *Bacillus subtilis* (BS). The compounds exhibited good antifungal activity against penicillium expansum (PE), *Aspergillus niger* (AN), *Trichoderma nigrum* (TL).

## RESULTS AND DISCUSSION

The target compounds 7(a-ziv) were synthesized through the route depicted in the scheme I. Disubstituted anthranilic acid reacts with R (a-f) to give respective substituted benzooxazin-4-one derivatives (2) were confirmed by 0.9  $\text{CH}_3$ , 7.5 – 8.1  $\text{C}_6\text{H}_5$ , 3.4  $\text{CH}_2\text{Cl}$   $^1\text{H-NMR}$  protons) Benzoyl glycine (3) reacts with H/ O-Cl/ O-OH/ P-( $\text{CH}_3$ )N/ P-O $\text{CH}_3$  substituted benzaldehyde R' (1-5) (4) to give respective substituted oxazoleone derivatives (5), (7.14 – 7.30  $\text{C}_6\text{H}_5$ , 7.3 – 7.6  $\text{C}_6\text{H}_5$ , 7.6  $\text{C}_6\text{H}_5\text{-CH}_2=$ , 5.0 OH, 2.85 P-( $\text{CH}_3$ )N, 3.73 O $\text{CH}_3$   $^1\text{H-NMR}$  protons) followed by treatment with ethylene diamine yielded corresponding substituted imidazole derivatives (6) (7.14 – 7.30  $\text{C}_6\text{H}_5$ , 7.3-7.6  $\text{C}_6\text{H}_5\text{-CH}_2=$ , 3.22  $\text{CH}_2$ , 2.91  $\text{CH}_2$ , 2.0  $\text{NH}_2$ , 7.6  $\text{C}_6\text{H}_5\text{-CH}_2=$ , 3.73 O $\text{CH}_3$ , 5.0 OH  $^1\text{H-NMR}$  protons), Finally condensation of (2) and (6) to give respective 7 (a-Ziv) (7.4 – 7.9  $\text{C}_6\text{H}_5$ , 3.46  $\text{CH}_2$ , 3.26  $\text{CH}_2$ , 7.3-7.26  $\text{C}_6\text{H}_5$ , 7.6  $\text{C}_6\text{H}_5\text{-CH}_2=$ , 7.14-7.3  $\text{C}_6\text{H}_5$ , 3.73 O $\text{CH}_3$ , 5.0 OH  $^1\text{H-NMR}$  protons) (120.9-147.1  $\text{C}_6\text{H}_5$ , 161.5 C=O, 164 CH, 46.9  $\text{CH}_2$ , 46.7  $\text{CH}_2$ , 46.6  $\text{CH}_2\text{Cl}$ , 166.6 C=O, 130.4 4-CH in imidazole, 146.5 CH in 2-  $\text{C}_6\text{H}_5$ , 126.1-130.2  $\text{C}_6\text{H}_5$  in imidazole, 108.4 CH 4-benzylidene, 115.8-158.3  $\text{C}_6\text{H}_5$  in 4-benzylidene, 55.9 O $\text{CH}_3$ , 40.3 P-( $\text{CH}_3$ )N  $^{13}\text{C-NMR}$  Protons) in moderate to good yields and this was confirmed by IR,  $^1\text{HNMR}$

and MS Spectral Studies. Among quinoxalines derivatives have registered good antibacterial activity against EC, PA, BS (Except 7b,7n,7ziv for EC, 7d,7o for PA, 7d,7q,7zi,7zii,7ziv for BS). These prepared quinoxalines have shown moderate to significant antifungal activity also against PE, AN, TL (Except 7a,7n,7o for PE, 7v,7ziv for AN, 7h,7p,7q,7x,7zii,7ziv for TL).

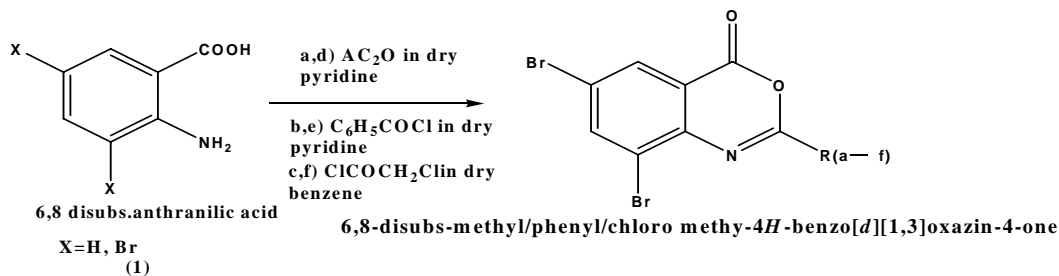
TABLE -1: PHYSICAL DATA

Compd(7)	X	R a-f	R' 1-5	Mol. Formula	M.W.	M.P.( <sup>0</sup> k)	YIELD (%)
a	H	CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	C <sub>27</sub> H <sub>22</sub> N <sub>4</sub> O <sub>2</sub>	434	907	67
b	H	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> N(C <sub>3</sub> ) <sub>2</sub>	C <sub>29</sub> H <sub>27</sub> N <sub>5</sub> O <sub>2</sub>	477	975	63
c	H	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>27</sub> H <sub>21</sub> ClN <sub>4</sub> O <sub>2</sub>	450	949	67
d	H	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> OH	C <sub>28</sub> H <sub>24</sub> N <sub>4</sub> O <sub>3</sub>	450	1019	94
e	H	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>28</sub> H <sub>24</sub> N <sub>4</sub> O <sub>3</sub>	464	953	75
f	H	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	C <sub>32</sub> H <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	496	990	72
g	H	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> N(CH <sub>3</sub> ) <sub>2</sub>	C <sub>34</sub> H <sub>29</sub> N <sub>5</sub> O <sub>2</sub>	539	1057	76
h	H	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>32</sub> H <sub>23</sub> ClN <sub>4</sub> O <sub>2</sub>	531	1032	75
i	H	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> OH	C <sub>32</sub> H <sub>24</sub> N <sub>4</sub> O <sub>3</sub>	512	1102	74
j	H	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>33</sub> H <sub>26</sub> N <sub>4</sub> O <sub>3</sub>	526	1036	78
k	H	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>5</sub>	C <sub>27</sub> H <sub>21</sub> ClN <sub>4</sub> O <sub>2</sub>	468	937	79
l	H	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> N(CH <sub>3</sub> ) <sub>2</sub>	C <sub>29</sub> H <sub>26</sub> ClN <sub>5</sub> O <sub>2</sub>	512	1005	74
m	H	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>27</sub> H <sub>20</sub> Cl <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	503	979	71
n	H	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> OH	C <sub>27</sub> H <sub>21</sub> ClN <sub>4</sub> O <sub>3</sub>	484	1049	75
o	H	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>28</sub> H <sub>23</sub> ClN <sub>4</sub> O <sub>3</sub>	498	983	73
p	Br	CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	C <sub>27</sub> H <sub>20</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	592	1052	79
q	Br	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> N(CH <sub>3</sub> ) <sub>2</sub>	C <sub>29</sub> H <sub>25</sub> Br <sub>2</sub> N <sub>5</sub> O <sub>2</sub>	635	1119	78
r	Br	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>27</sub> H <sub>19</sub> Br <sub>2</sub> ClN <sub>4</sub> O <sub>2</sub>	626	1094	75
s	Br	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> OH	C <sub>27</sub> H <sub>20</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>3</sub>	608	1163	84
t	Br	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>28</sub> H <sub>22</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>3</sub>	622	1134	84
u	Br	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	C <sub>28</sub> H <sub>22</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	654	1098	85
v	Br	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> N(CH <sub>3</sub> ) <sub>2</sub>	C <sub>32</sub> H <sub>27</sub> Br <sub>2</sub> N <sub>5</sub> O <sub>2</sub>	697	1202	94
w	Br	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>32</sub> H <sub>21</sub> Br <sub>2</sub> ClN <sub>4</sub> O <sub>2</sub>	688	1177	94
x	Br	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> OH	C <sub>32</sub> H <sub>22</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>3</sub>	670	1246	94
y	Br	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>33</sub> H <sub>24</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>3</sub>	684	1180	91
z	Br	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>5</sub>	C <sub>27</sub> H <sub>19</sub> Br <sub>2</sub> ClN <sub>5</sub> O <sub>2</sub>	626	1082	68
zi	Br	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> (CH <sub>3</sub> ) <sub>2</sub>	C <sub>29</sub> H <sub>24</sub> Br <sub>2</sub> ClN <sub>5</sub> O <sub>2</sub>	669	1149	64
zii	Br	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>27</sub> H <sub>18</sub> Br <sub>2</sub> Cl <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	661	1124	61
ziii	Br	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> OH	C <sub>27</sub> H <sub>19</sub> Br <sub>2</sub> ClN <sub>4</sub> O <sub>3</sub>	642	1193	68
ziv	Br	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>28</sub> H <sub>21</sub> Br <sub>2</sub> ClN <sub>4</sub> O <sub>3</sub>	656	1128	67

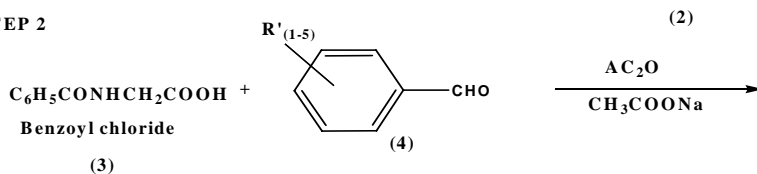
TABLE-2: SPECTARL DATA

Compd (7)	X	R a-f	R' 1-5	Mol.Formula	M+1	<sup>1</sup> H- NMR	IR CM <sup>-1</sup>
a	H	CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	C <sub>27</sub> H <sub>22</sub> N <sub>4</sub> O <sub>2</sub>	434	3.26, 3.46, 7.6, 0.9, 7.3 - 7.6	3200, 1450, 1650
b	H	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> N(C <sub>3</sub> ) <sub>2</sub>	C <sub>29</sub> H <sub>27</sub> N <sub>5</sub> O <sub>2</sub>	477	3.26, 3.46, 7.6, 0.9, 2.82	3200, 3400, 1680
c	H	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>27</sub> H <sub>21</sub> ClN <sub>4</sub> O <sub>2</sub>	450	3.26, 3.46, 7.6, 0.9, 7.3 - 7.6	3250, 1480, 1650
d	H	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> OH	C <sub>28</sub> H <sub>24</sub> N <sub>4</sub> O <sub>3</sub>	450	3.26, 3.46, 7.6, 0.9, 5.0	3400, 2900, 1700
e	H	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>28</sub> H <sub>24</sub> N <sub>4</sub> O <sub>3</sub>	464	3.26, 3.46, 7.6, 0.9, 3.70	3000, 1510, 1680
f	H	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	C <sub>32</sub> H <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	496	3.24, 3.46, 7.6, 7.3 - 7.6	3200, 2900, 1680
g	H	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> N(CH <sub>3</sub> ) <sub>2</sub>	C <sub>34</sub> H <sub>29</sub> N <sub>5</sub> O <sub>2</sub>	539	3.24, 3.46, 7.6, 7.4, 2.85	3200, 3420, 1650
h	H	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>32</sub> H <sub>23</sub> ClN <sub>4</sub> O <sub>2</sub>	531	3.24, 3.46, 7.6, 7.4, 7.4 - 7.6	3000, 1450, 1650
i	H	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> OH	C <sub>32</sub> H <sub>24</sub> N <sub>4</sub> O <sub>3</sub>	512	3.24, 3.46, 7.6, 7.3, 5.05	3400, 2900, 1670
j	H	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>33</sub> H <sub>26</sub> N <sub>4</sub> O <sub>3</sub>	526	3.24, 3.46, 7.6, 7.3 - 7.6, 3.73	3100, 1470, 1640
k	H	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>5</sub>	C <sub>27</sub> H <sub>21</sub> ClN <sub>4</sub> O <sub>2</sub>	468	3.26, 3.46, 7.6, 3.4, 7.3 - 7.6	3150, 1500, 1650
l	H	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> N(CH <sub>3</sub> ) <sub>2</sub>	C <sub>29</sub> H <sub>26</sub> ClN <sub>5</sub> O <sub>2</sub>	512	3.26, 3.44, 7.6, 3.4, 2.83	3420, 1520, 1680
m	H	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>27</sub> H <sub>20</sub> Cl <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	503	3.23, 3.46, 7.6, 3.4, 7.2 - 7.6	2900, 3000, 1680
n	H	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> OH	C <sub>27</sub> H <sub>21</sub> ClN <sub>4</sub> O <sub>3</sub>	484	3.20, 3.46, 7.6, 3.4 - 5.50	3400, 2920, 1650
o	H	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>28</sub> H <sub>23</sub> ClN <sub>4</sub> O <sub>3</sub>	498	3.26, 3.50, 7.6, 3.29, 3.76	3200, 1300, 1680
p	Br	CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	C <sub>27</sub> H <sub>20</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	592	3.26, 3.46, 7.6, 0.9, 7.3 - 7.56	3100, 2920, 1650
q	Br	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> N(CH <sub>3</sub> ) <sub>2</sub>	C <sub>29</sub> H <sub>25</sub> Br <sub>2</sub> N <sub>5</sub> O <sub>2</sub>	635	3.26, 3.46, 7.6, 0.9, 2.80	3200, 1400, 3400
r	Br	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>27</sub> H <sub>19</sub> Br <sub>2</sub> ClN <sub>4</sub> O <sub>2</sub>	626	3.26, 3.45, 7.6, 0.9, 6.68, 7.13	2900, 1470, 1680
s	Br	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> OH	C <sub>27</sub> H <sub>20</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>3</sub>	608	3.26, 3.46, 7.6, 0.9, 5.05	2920, 1450, 1700
t	Br	CH <sub>3</sub>	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>28</sub> H <sub>22</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>3</sub>	622	3.26, 3.46, 7.6, 0.9, 3.77	2100, 1500, 1680
u	Br	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	C <sub>28</sub> H <sub>22</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	654	3.26, 3.46, 7.6, 7.4, 7.3 - 7.6	3200, 1600, 1700
v	Br	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> N(CH <sub>3</sub> ) <sub>2</sub>	C <sub>32</sub> H <sub>27</sub> Br <sub>2</sub> N <sub>5</sub> O <sub>2</sub>	697	3.26, 3.46, 7.6, 7.45 - 7.6, 2.79	3380, 1450, 1650
w	Br	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>32</sub> H <sub>21</sub> Br <sub>2</sub> ClN <sub>4</sub> O <sub>2</sub>	688	3.26, 3.46, 7.6, 6.98, 7.22	3200, 1400, 1700
x	Br	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> OH	C <sub>32</sub> H <sub>22</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>3</sub>	670	3.26, 3.46, 7.6, 7.3 - 7.6, 5.0	3500, 1450, 1720
y	Br	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>33</sub> H <sub>24</sub> Br <sub>2</sub> N <sub>4</sub> O <sub>3</sub>	684	3.26, 3.46, 7.6, 7.3 - 7.6, 3.77	2900, 1470, 1650
z	Br	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>5</sub>	C <sub>27</sub> H <sub>19</sub> Br <sub>2</sub> ClN <sub>5</sub> O <sub>2</sub>	626	3.26, 3.46, 7.6, 3.44, 7.4 - 7.56	2900, 1450, 1700
zi	Br	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> N(CH <sub>3</sub> ) <sub>2</sub>	C <sub>29</sub> H <sub>24</sub> Br <sub>2</sub> ClN <sub>5</sub> O <sub>2</sub>	669	3.26, 3.46, 7.6, 3.40, 2.88	3400, 1500, 1650
zii	Br	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> Cl	C <sub>27</sub> H <sub>18</sub> Br <sub>2</sub> Cl <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	661	3.26, 3.46, 7.6, 3.45, 7.2 - 7.6	3200, 1450, 1650
ziii	Br	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> OH	C <sub>27</sub> H <sub>19</sub> Br <sub>2</sub> ClN <sub>4</sub> O <sub>3</sub>	642	3.26, 3.46, 7.6, 3.40, 7.3 - 7.6	3450, 1450, 1700
ziv	Br	CH <sub>2</sub> Cl	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	C <sub>28</sub> H <sub>21</sub> Br <sub>2</sub> ClN <sub>4</sub> O <sub>3</sub>	656	3.26, 3.46, 7.6, 3.42, 3.70	2900, 1470, 1650

SCHEME 1  
STEP 1



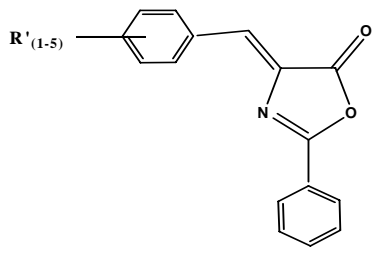
STEP 2



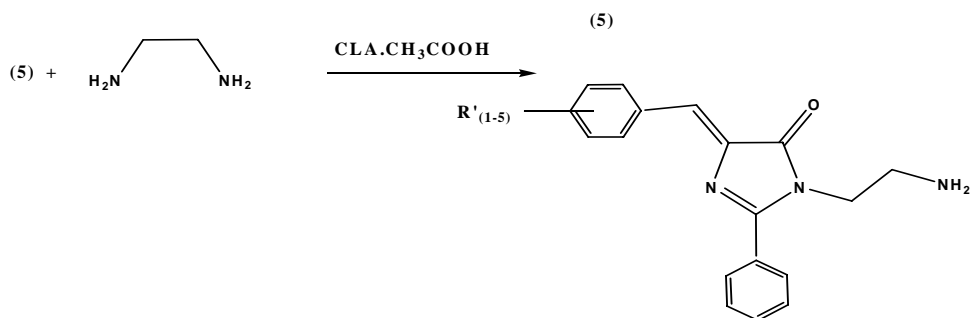
R'

---

R'<sub>1</sub> = H  
R'<sub>2</sub> = (CH<sub>3</sub>)N  
R'<sub>3</sub> = O-Cl  
R'<sub>4</sub> = O-OH  
R'<sub>5</sub> = P-OCH<sub>3</sub>

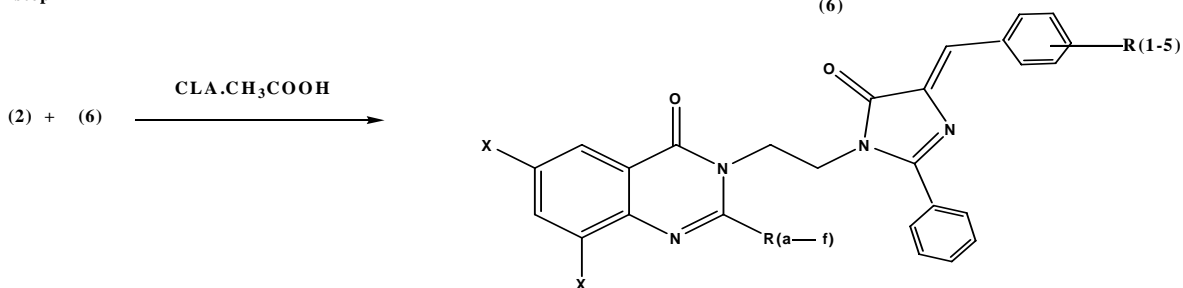


step 3



(4Z)-1-(2-aminoethyl)-4-subst. benzylidene-2-phenyl-1H-imidazol-5(4H)-one

step 4



3-(2-((19Z)-4subst.-benzylidene-4,5-dihydro-5-oxo-2-phenylimidazol-1-yl)ethyl)-6,8-disubs-2-methylquinazolin-4(3H)-one  
(7)

TABLE-3: ANTIMICROBIAL ACTIVITY

Compd(7)	Anti Bacterial (25 µg / disc)			Anti Fungal (1000 µg / disc)		
	Zone of Inhibition in mm			Zone of Inhibition in mm		
	EC	PA	BS	PE	AN	TL
a	14	16	12	-	67	70
b	-	13	12	83	84	85
c	12	13	14	75	83	84
d	14	-	-	80	81	80
e	13	13	12	65	79	85
f	14	14	13	79	77	85
g	16	12	18	80	75	85
h	12	13	13	80	77	-
i	16	14	12	73	78	79
j	15	18	16	63	79	81
k	11	12	16	81	81	83
l	14	13	13	76	82	84
m	12	16	12	73	83	80
n	-	14	12	-	84	83
o	10	-	14	-	85	85
p	12	13	13	71	86	-
q	13	12	-	63	83	-
r	14	16	12	58	73	84
s	16	12	12	83	72	85
t	14	14	12	81	59	83
u	14	12	13	82	58	80
v	13	16	12	70	-	83
w	11	13	12	71	63	85
x	16	16	13	79	63	-
y	12	14	10	63	77	85
z	18	14	8	62	71	83
zi	18	14	-	65	63	85
zii	18	13	-	68	68	-
ziii	16	14	12	69	69	-
ziv	-	14	-	71	-	79
STD	14	12	20	93	92	90

## REFERENCES

1. D.L.Trepanier and S.Sunder, US.Pat., 3,919,220 (C.A.,1976, 84 121907u)
2. R.A.LeMahinu, M.Carson, A.F.Weiton,H.W.Baruth and B.Yaremko, *J.Med.Chem.*,**26**, 107, (1983).
3. R.W.Kierstead and J.W.Tilley, Can.Pat. 1, 189, 509 (C.A., 1986, 104,
4. A.A.Khalil, S.G.Addel Hamide, A.M.Al-Obaid, H.T.El-Subbagh, *Arch. Pharm. Pharm.Med. Chem.*, **336**, 95-103 (2003).
5. Sarah T. Al-Rashood, Ihsan A. Aboldahab, Mahmoud N. Nagi, LailaA. Aboouzeid,AlaaA.M.Abdel-Aziz, Sami G.Abdel-Hamide, Khairia M. Youssef, Addulrahman M. Al-Obaid and Hussein I. El-Subbagh, *Bioorg. Med.Chem.*, **14**, 8608-8621 (2006).03 (2003).
6. V.Alagarsamy, V.Rajasolomon and K.Dhanabal, *Bioorg.Med.Chem*,**15**, 235-241 (2007).
7. J.L.Marrosz, B. Duszynska, S.Charakchieva-minol, A.J.Bojarski, M.j.mokrosz,R.L.Wrra,L.Janda,L.Strekowski, *Eur.J.Med.Chem.*, **31**, 973-980 (1996).
8. Neha Dixid,V.K.Salvi, G.L.Talesara, *Ind.J.Chem.Sci.* , **3(1)**, 31-36 (2005).
9. M.Shah, P.Patel, S.Korgaokav and H.Parekh, *Indian J.Chem.*, **35B**, 1282(1996).
10. D.G.Kamble, B.Yadav and S.S.Sangapura, *Indian J.Chem.* , **9**, 25 (1999).
11. R.Suthakaran, G.Nagarajan, V.Balasubramaniam, K.Suganthi and G.Velrajan, *Indian J.Heterocycl. Chem.*, **4**, 201 (2004).
12. R.Suthakaran, G.Somasekhar,Ch.Sridevi, M.Marikannan, K.Suganthi and G.Nagarajan, *AsianJ.Chem*, **19(5)**, 3353-3362 (2007).
13. Y.Rajendra prasad, A.lakhmana rao,P.Ravi kumar, N.Koteswara Rao and B,Ganga Rao,*Int.J.Pharmco.Bio.Sci*, **1(2)**,1-4 (2007).

(Received: 8 October 2007

Accepted: 4 January 2008

RJC-134)